QUALITY OF SURFACE WATERS IN WILTON, CONNECTICUT

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Multiply inch-pound unit	<u>By</u>	To obtain SI units
foot (ft)	0.3048	meter (m)
mile (mi)	1.609 2.590	kilometer (km)
square mile (mi <sup>2</sup> ) acre-foot (acre-ft)	1,233	square kilòmeters (km²) cubic meter (m³)
ton (short)	0.9072	megagram (Mg)
cubic foot per	0.02832	cubic meter per
second (ft <sup>3</sup> /s)	0,0002	second (m <sup>3</sup> /s)
micromho per	1.000	microsiemen per
centimeter (umho/cm)	2000	centimeter (uS/cm)
degree Fahrenheit (°F)	$^{\circ}C = (^{\circ}F-32)/1.8$	degree Celcius (°C)

### QUALITY OF SURFACE WATERS IN WILTON, CONNECTICUT

Ву

### Kenneth P. Kulp

### **ABSTRACT**

Water, bed material, and biological samples were collected and analyzed at 10 surface-water gaging sites on six streams in the Town of Wilton, Connecticut over a 2-year period. The data indicate fair to excellent water quality. Fecal coliform bacteria, pH, alkalinity, iron, and manganese are the parameters that most often exceed recommended limits established by either the U.S. Environmental Protection Agency or the Connecticut Department of Environmental Protection.

Data from sites on the Norwalk and East Branch Silvermine Rivers indicate little if any undesirable changes in water quality take place as they flow through the study area.

### INTRODUCTION

In April 1976, the U.S. Geological Survey, in cooperation with the Town of Wilton, Conn., began a program of collecting water-quality data. The objectives of the program were to determine baseline water-quality data of the major rivers and streams, and to evaluate downstream changes in the quality of the Norwalk and Silvermine Rivers.

Data collection ended in April 1978. This report summarizes the program and presents all the data collected.

### DESCRIPTION OF THE STUDY AREA

The Town of Wilton is in the southwest corner of Connecticut, north of the city of Norwalk, in Fairfield County (fig. 1). The town is predominantly suburban, with some light industry. In recent years, the area has undergone rapid development, which will probably continue.

Based on records of the Norwalk River at South Wilton, streamflows for the 1976 water year (Oct. 1, 1975 - Sept. 30, 1976) were approximately 70 percent above average. Streamflows in the 1977 water year were 93 percent of average and in the 1978 water year were 55 percent of average.

Ten sites were sampled. Their locations are shown on figure 1, and other information is given in table 1.

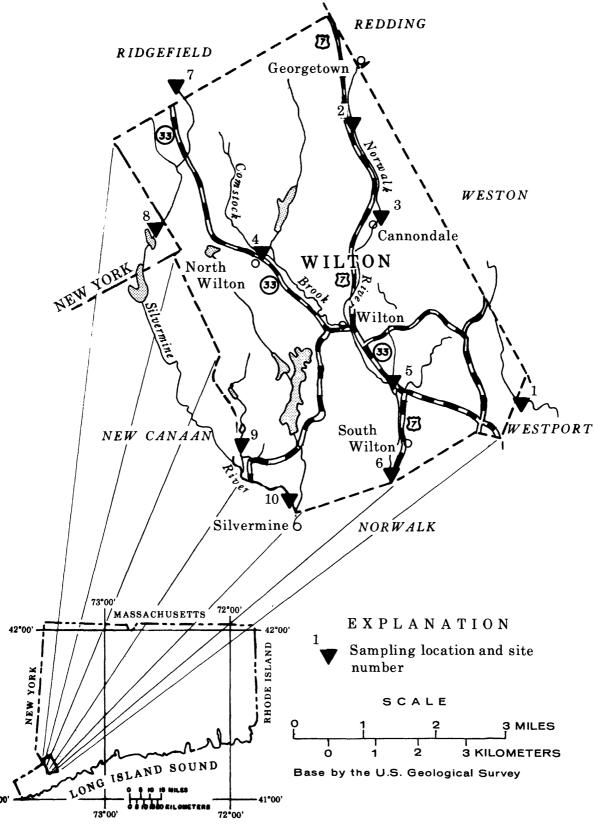


FIGURE 1.-Study area and sampling locations.

Table 1.--Sampling site identification, location, and drainage area

Pro- ject	usgs		Coordinate		Drainage	
No.	station No.	Name	Latitude	Longitude	area (mi²)	Location
1	01209400	West Branch Saugatuck River near Weston, Conn.	41°10'41"	73°22'50"	10.8	At Newton Turnpike bridge, 1.5 miles south of Weston, Conn.
2	01209570	Norwalk River at Georgetown, Conn.	41°14'45"	73°26'05"	14.5	At bridge on Old Mill Road, O.8 mile south of Georgetown, Conn
3	01209572	Norwalk River at Cannondale, Conn.	41°13'52"	73°25'35"	15.2	At bridge on Honey Hill Road, 1 mile north of Cannondale, Conn.
4	01209600	Comstock Brook at North Wilton, Conn.	41°12'45"	73°27'31"	3.53	At bridge on Nod Hill Road, at North Wilton, Conn.
5	01209680	Bryant Brook at Wilton, Conn.	41°11'01"	73°25'09"	0.99	At U.S. Route 7 bridge, 1 mile southeast of Wilton, Conn.
δ	01209700	Norwalk River at South Wilton, Conn.	41°09'49"	73°25'11"	30.0	At bridge on Kent Road, at South Wilton, Conn.
7	01209715	East Branch Silvermine River near Ridgefield, Conn.	41°14'59"	73°29'03"	1.36	At Gay Road bridge, 2.5 miles southeast of Ridgefield, Conn.
8	01209716	East Branch Silvermine River near North Wilton, Conn.	41°13'08"	73°29'28"	2.64	At Silver Spring Road bridge by Browns Reservoir, N.Y., 1.5 miles northwest of North Wilton, Conn.
9	01209730	Thayers Brook near Silvermine, Conn.	41°10'10"	73°27'51"	1.59	At Huckleberry Hill Road bridge, below Domenicks Pond, 1.5 miles northwest of Silvermine, Conn.
0	01209734	Silvermine River near Silvermine, Conn.	41°09'25"	73°27'01"	16.0	At bridge on private road, 0.5 mile north of Silvermine, Conn

### Stream Systems

Three rivers flow through Wilton. The principal river is the Norwalk, with a drainage area of  $30.0~\text{mi}^2$  (77.7 km²) at South Wilton; other rivers are the Silvermine, drainage area of  $16.0~\text{mi}^2$  ( $41.4~\text{km}^2$ ) near Silvermine, and the West Branch Saugatuck, drainage area of  $10.8~\text{mi}^2$  ( $28~\text{km}^2$ ) at Weston. The Silvermine River flows into the Norwalk River at Norwalk, just before it enters Long Island Sound. The West Branch Saugatuck River joins the Saugatuck River north of Weston, and flows into Long Island Sound at Saugatuck. Significant small streams are Comstock Brook and Bryant Brook, tributaries to the Norwalk River, and Thayers Brook, a tributary to the Silvermine River.

### Geology

The study area is underlain by gneiss and other metamorphic and igneous rocks (Kroll, 1977). The bedrock is discontinuously mantled by unconsolidated deposits, primarily of glacial origin. The most widespread unconsolidated deposits are till, that occurs in upland areas and stratified drift, that is generally confined to major valleys (Ryder and others, 1970).

### METHODS AND TECHNIQUES

During the initial phase of the program, April to June 1976, each of the 10 sites was sampled monthly. Subsequent sampling was restricted to the three Norwalk River sites. These sites were sampled approximately every month through November 1977 and in March and April 1978.

The water at each site was analyzed in the field for temperature, specific conductance, pH, dissolved oxygen, total and fecal coliform, and fecal streptococci bacteria. Samples were also collected and analyzed for selected chemical constituents. Streamflow was measured at each site when samples were collected. The methods and techniques used to collect and analyze the samples are described by Brown and others (1970), Greeson and others (1977), and Buchanan and Somers (1969).

In addition, bed material and biological studies were carried out at sites 2, 3, and 4 in June 1976. Bed material samples were analyzed for metals and pesticides. Samples of suspended or floating plant life (phytoplankton), algae attached to submerged substrates (periphyton), and bottom dwelling animals (benthic invertebrates) were collected and analyzed for numbers, types, and community structure. The water was also analyzed for algal growth potential (AGP), a measure of its ability to support plant life.

A summary of the sampling frequency at each site is given in table 2.

Table 2.--Sampling frequency at sites

Site No.	Time period	Number of water samples	Number of bed material and biological sample
1	Apr. 1976-June 1976	3	
2	Apr. 1976-Apr. 1978	22	1
3	Apr. 1976-Apr. 1978	22	1
4	Apr. 1976-June 1976	3	1
5	Apr. 1976-June 1976	3	
6	Apr. 1976-Apr. 1978	22	
7	Apr. 1976-June 1976	3	
8	Apr. 1976-June 1976	3	
9	Apr. 1976-June 1976	3	<del></del>
10	Apr. 1976-June 1976	3	

### WATER QUALITY

Results of the water sampling program have been evaluated on a site-by-site basis, using the criteria in table 3, which indicate limiting values of various chemical constituents and physical properties of water based on the protection of water users, aquatic organisms, and aesthetic value.

Additional information concerning the sources and significance of selected chemical, physical, and biological properties of water determined in this study are contained in table 4. The results of the bed material and biological studies are discussed separately, as they were not performed at all sites.

All of the water-quality data collected are given in table 11. For most sites, data collected have been compared with Connecticut Water Quality Standards and Classifications (Connecticut Department of Environmental Protection, 1977).

The official State classification of the waters investigated in this study are shown in figure 2, and a condensed list of the classification standards is shown in table 5. (For more detailed information on the classification of various reaches of streams and the State classification standards, refer to the cited publication.) Because all parameters contained in the State's quality standards were not determined in this study and the number of analyses for any site is limited, it is not possible to assess how the water should be classified according to the State standards.

Table 3. -- Selected water-quality criteria

Chemical constituent or physical property	Limiting value	Use	Basis for selection
Alkalinity, total (as CaCO <sub>3</sub> )	å20 mg∕L	1	A
Chloride	250 mg/L	4	C
Coliform Bacteria, Fecal	b20 col/100 mL c200 col/100 mL b1,000 col/100 mL	3 6 5	D A,D D
Color	d <sub>15</sub> color units d <sub>75</sub> color units	4 2	C A
Copper	1,000 ug/L	2,4	A,C
Iron	300 ug/L 1,000 ug/L	2,4	A,C A
Lead, dissolved	50 ug/L	2,3	A,B
Manganese	50 ug/L	2,4	A,C
Methylene Blue Active Substance	0.5 mg/L	4	С
Nitrate (as N)	10 mg/L	2,3	A,B
Nitrite (as N)	1 mg/L	2	A
Oxygen, dissolved	a5 mg/L a4 mg/L	1 6	A,D D
рН	5.0-9.0 6.0-8.5 6.5-8.0 6.5-8.5 6.5-9.0	2 1,6 5 4 1	A D D A,C A
Sodium	20 mg/L	3	D
Solids, dissolved	500 mg/L	4	С
Sulfate	250 mg/L	4	С
Turbidity	<sup>e</sup> 25 Jackson turbidity units (JTU)	1	D
Zinc	5,000 ug/L	2,4	A,C

### Water Use and(or) for the Protection of:

- 1. Freshwater aquatic life
- 2. Domestic water supply source
- 3. Potable drinking water, based on health effects
- 4. Potable drinking water, based on aesthetic considerations
- 5. Primary contact 6. Aesthetic value

### Basis for Selection

- A. Maximum levels recommended by: Quality Criteria for Water, 1977, U.S.
- A. Maximum revers recommended by: Quarity Criteria for Water, 1977, 0.5.

  Environmental Protection Agency.

  B. Maximum contaminant level established by: National Interim Primary Drinking Water Regulations 1975, U.S. Environmental Protection Agency.

  C. Maximum contaminant level recommended for the Proposed National Secondary Drinking Water Regulations, 1977, U.S. Environmental Protection Agency.

  D. Maximum levels established by: State of Connecticut Water Quality Standards, 1980, Connecticut Department of Environmental Protection (see Table 5).

- $\underline{a}$ / Minimum recommended value.  $\underline{b}$ / Arithmetic mean.  $\underline{c}$ / Log mean, based on a minimum of five samples.  $\underline{d}$ / As determined by use of the platinum-cobalt method (Standard Methods, 1971).
- e/ Unit of measurement based on the Jackson candle turbidimeter.

Chemical constituent or physical property	Source	Significance
Alkalinity	The capacity to neutralize acid, which is usually produced by dissolved carbonates, bicarbonates, and(or) hydroxides in the water.	The buffering capacity of water relates to pH, which has direct and indirect effects on aquatic life, and other uses of water.
Calcium (Ca) and magnesium (Mg)	Dissolved from rocks and soils, especially those containing calcium silicates, clay minerals, and carbonate lenses.	Hardness and scale-forming properties of water ar caused principally by dissolved bicarbonates and sulfates of calcium and magnesium. (See hardness.) Hard water is objectionable for electroplating, tanning, dyeing and textile processing. It also causes scale formation in steam boilers, water heaters, and pipes.
Carbon (C) organic	Derived from organic material.	Indicative of organic pollution loads
Carbon dioxide (CO <sub>2</sub> )	Derived from atmosphere, respiration of aquatic organisms, decomposition of organic matter, and breakdown of bicarbonate due to acids.	Can have detrimental effects on fish, and contri- butes to the corrosiveness of the water
Carbonate (CO3) and bicarbonate (HCO3)	Dissolved from carbonate and calcium silicate minerals by reaction with carbon dioxide in water. Decaying vegetation, sewage, and industrial wastes are also important sources.	Carbonates of calcium and magnesium cause hard- ness, form scale in boilers and pipes, and releas corrosive carbon dioxide gas. (See hardness.) Water of low mineral content and low bicarbonate content in proportion to carbon dioxide is acidic and corrosive.
Chloride (Cl)	Dissolved from rocks and soils in small amounts. Other sources are animal wastes, sewage, road salt, industrial wastes, and seawater.	Large amounts in combination with calcium will result in a corrosive solution and in combination with sodium will give water a salty taste.
Coliform, total	A group of bacteria widely distributed in the environment, some types live in soil, others live in the intestinal tract of warm blooded animals.	As a group are generally considered harmless. Coliforms in water are considered indicators of possible warm-blooded animal waste contamination, including that of humans.
Coliform, fecal	Type of coliform bacteria which live in the intestinal tract of warm blooded animals, including humans. Enter water by way of fecal wastes of these animals.	Presence in water specifically indicates fecal waste contamination by warm blooded animals.
Color	May be imparted by iron and manganese compounds, algae, weeds, and humus. May also be caused by inorganic or organic wastes from industry. True color of water is considered to be only that remaining in solution after the suspended material has been removed.	Water for domestic and some industrial uses shoul be free of perceptible color. Color in water is objectionable in food and beverage processing and many manufacturing processes.
Copper (Cu)	Presence in more than trace concentrations is usually derived from corrosion of copper pipes, industrial wastes, and copper salts used as algacides and herbicides.	Imparts an objectionable taste to water, toxic to some aquatic organisms. Undesirable in water for some industrial uses.
Hardness	Primarily due to calcium and magnesium, and to a lesser extent to iron, manganese, aluminum, barium, and strontium. There are two classes of hardness: carbonate (temporary) and noncarbonate (permanent). Carbonate hardness refers to the hardness balanced by equivalents of carbonate and bicarbonate ions; noncarbonate refers to the remainder of the hardness. Most waters in the basin are classified as soft to moderately hard.	Hard water uses more soap to lather and deposits soap curds on bathtubs. Water having a hardness of more than 120 mg/L is commonly softened for domestic use. Hardness forms scale in boilers, water heaters, radiators, and pipes, causing a decrease in rate of heat transfer and restricted flow of water. In contrast, water having a very low hardness may be corrosive.
Iran (Fe)	Dissolved from minerals that contain oxides, sulfides, and carbonates of iron. Decaying vegetation, iron objects that are in contact with water, sewage, and industrial waste are also major sources.	On exposure to air, iron in ground water oxidizes to a reddish-brown precipitate. More than about 300 ug/L stains laundry and utensils, causes unpleasant odors, and favors growth of iron bacteria. Iron in water is objectionable for food and textile processing. Most iron-bearing water, when treated by aeration and filtration, are satisfactory for domestic use.
Lead (Pb)	Atmospheric deposition, erosion, and leaching of soil and rocks, and municipal and industrial waste.	A cumulative poison to animals, including man.
Manganese (Mn)	Dissolved from many rocks and soils. Commonly associated with iron in natural waters but less common.	More than 5D ug/L oxidizes to a black precipitate. Manganese has the same undesirable characteristics as iron but is more difficult to remove
Methylene blue active substance (MBAS)	A measure of the concentrations of apparent detergents in water. The most common sources are synthetic household detergent residues in sewage and waste waters.	High concentration of detergents cause undesirable taste, foaming, and odors. Indicates possible presence of sewage or industrial waste. Low values of MBAS may also be reflective of phenols, proteins, inorganic chlorides, cyanates, nitrates thiocyanates, and organic compounds having amine groups, which also give positive results to the MBAS determination. Some of these substances may

or physical property	Source	Significance
Nitrogen (N)	Derived from the atmosphere, leaching from rocks, and soils, decomposition of organic materials, and various biological processes. Also from fertilizers, municipal and industrial wastes and sewage.	Combines with other elements and at various oxidation states to form chemical species such as ammonia, nitrate, and nitrite.
Ammonia (NH4)	A degradation product of nitrogenous organic material. Also derived from industrial wastes.	Indicative or organic pollution and toxic to aquatic organisms.
Nitrate (NO3)	Sewage, industrial wastes, fertilizers, and decaying vegetation are major sources. Lesser amounts are derived from precipitation and solution processes.	Small amounts have no affect on usefulness of water A concentration greater than 10 mg/L generally indicates pollution. Nitrate encourages growth of alga and other organisms which produce undesirable taste and odors. Water containing more than 44 mg/L has reportedly caused methemoglobemia, which is often fatal to infants (Comly, 1945). Recommended maximu 10 mg/L of nitrate expressed as N, which is equivalent to 44 mg/L nitrate expressed as NO3.
Nitrite (NO <sub>2</sub> )	An intermediate product of the nitrification process. Formed from the nitrate or ammonium ions by micro-organisms found in water, soil, sewage, and animal digestive tracts.	Very unstable in the presence of oxygen, therefore it is found in minute quantities in most natural waters. It is sometimes indicative of organic pollution.
Oxygen, dissolved (D.O.)	Derived from the atmosphere and from photosynthesis by aquatic vegetation. Amount varies with temperature and pressure and decreases during breakdown of waste material. Concentration can be expressed in mg/L or as a percentage of saturation.	Dissolved oxygen in surface water is necessary for support of fish and other aquatic life. It causes precipitation of iron and manganese in well water and can cause corrosion of metals.
рН	Water having concentrations of acids, acid- generating salts, and free carbon dioxide has a low pH. Where carbonates, bicarbonates, hydrox- ides, phosphates and silicates are dominant, the pH is high. Most natural waters range between 6 and 8.	A pH of 7.0 indicates neutrality of a solution. Values higher than 7.0 denote alkaline characteristics; values lower than 7.0 indicate acid characteristics. Acid waters and excessively alkaline waters corrode metals.
Phosphorus (P)	Leaching of soils and rocks, fertilizer, decomposition of plants and animals, sewage, and industrial effluents.	In the elemental form, it is toxic. As phosphate, it is an essential plant nutrient, often being the limiting factor in plant growth.
Silica (SiO <sub>2</sub> )	Dissolved from practically all rocks and soils.	High concentrations precipitate as hard scale in boilers, water heaters, and pipes. Inhibits deterioration of zeolite-type softeners and corrosion of iron pipes.
Sodium (Na)	Dissolved from practically all rocks and soils. Sewage, industrial wastes, road salt, and sea water are also major sources. Most home water softeners increase the amount of sodium in water by exchanging it for calcium and magnesium.	More than 50 mg/L may cause foaming in steam boilers. The maximum permitted for people restricted to a low salt diet is 20 mg/L.
Solids, dissolved	Includes all dissolved mineral constituents de- rived from solution of rocks and soils. Locally augmented by mineral matter in sewage and indus- trial wastes. Measured as residue on evaporation at 180° or calculated as numerical sum of amounts of individual constituents.	Water containing more than 1,000 mg/L dissolved solids is undesirable for public and private supplies and most industrial purposes.
Specific conductance	Specific conductance, or the capacity of water to conduct an electric current. Reflects geochemistry of basin, also affected by pollution.	Relates approximately to dissolved-solids content. In the Norwalk River, 0.062 times the specific conductance is approximately equivalent to the dissolved-solids concentration.
Streptococci, fecal	A group of bacteria which live in the intestinal tract of warm blooded animals. Enter water with animal fecal matter.	Presence of fecal streptococci bacteria specifically indicates fecal waste contamination. These bacteria grow in the intestinal tract in different proportions to fecal coliform bacteria in various animals, thus they are useful in determining the source of fecal contamination.
Sulfate (SO <sub>4</sub> )	Dissolved from rocks and soils containing sulfur compounds, especially iron sulfide; also from sulfur compounds dissolved in precipitation and sewage and industrial wastes.	Sulfates of calcium and magnesium cause permanent hardness and form hard scale in boilers and hot water pipes.

Chemical constituent or physical property	Source	Significance		
Temperature	Fluctuates seasonally in streams and shallow aquifers. At depths greater than 30 feet, ground-water temperature remains within 2°C or 3°C of mean annual air temperature [10° to 11°C for the report area (Ryder and others, 1970). Disposal of water used for cooling or industrial processing may cause local temperature anomalies.	Affects the usefulness of water for many purposes. For most uses, especially cooling, water of uniformly low temperatures is desired. A rise of a few degrees in the temperature of a stream may limit its capacity to support aquatic life. Warm water carries less oxygen in solution and is more corrosive than cold water.		
Turbidity	An optical property of water attributed to suspended or colloidal matter which inhibits light penetration. May be caused by microorganisms, algae, suspended mineral substances including iron and manganese compounds, clay, silt, sawdust, fibers, or other materials. May result from natural processes of erosion or from the addition of domestic sewage, wastes from industries such as pulp and paper manufacturing, or sediment from construction activities.	Excessive concentrations are harmful or lethal to fish and other aquatic life. Turbidity is also undesirable in water used by most industries, especially in process water. Turbidity can modify water temperature.		
Zinc	Dissolved from rocks and ores which are slightly soluble, more so in acid water. Chlorides and sulfates of zinc are highly soluble. Also derived from industrial wastes.	Toxic to various aquatic organisms. High concentrations are detrimental to some crops.		

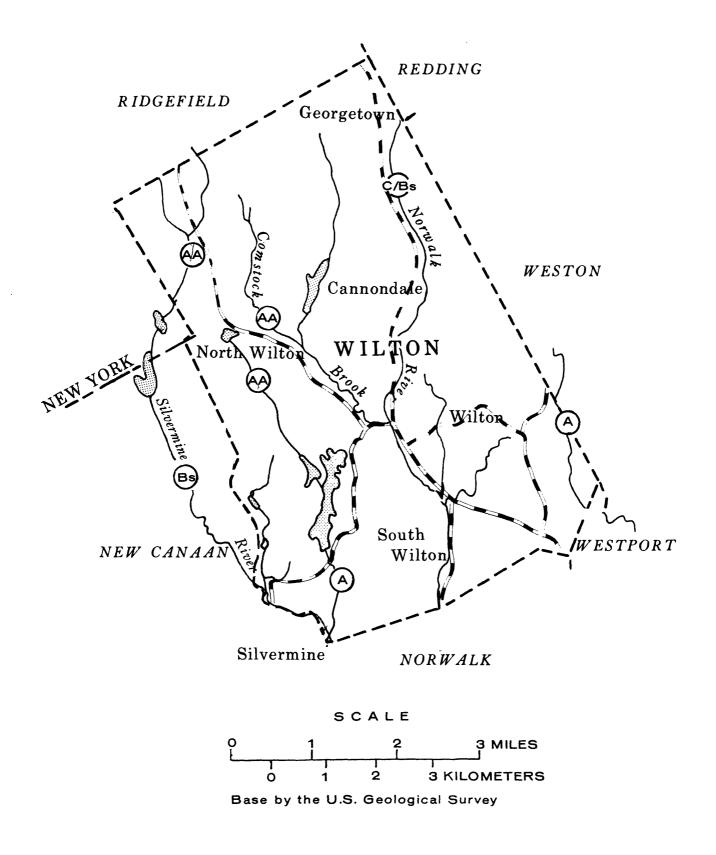


FIGURE 2.-Classification of surface waters in the study area.

### Table 5,--Classification standards for Connecticut inland waters

[Modified from Connecticut Water Quality Standards and Classifications, Connecticut Department of Environmental Protection 1977]

	Class AA	Class A	Class B	Class C
Use	Existing or proposed drinking water-supply impoundments and tributary surface waters.	May be suitable for drinking water supply and/or bathing; suitable for all other water uses; character uniformly excellent.	Suitable for bathing, other recreational purposes, agricultural uses, certain industrial processes and cooling; excellent fish and wildlife habitat; good aesthetic value.	Suitable for fish and wildlife habitat, recreational boating, and certain industrial processes and cooling; good aesthetic value.
Olssolved oxygen	Not less than 5 mg/L at any time	Not less than 5 mg/L at any time.	Not less than 5 mg/L at any time.	Not less than 4 mg/L at any time.
Color and turbidity [Jackson turbidity units (JTU)]	Turbidity shall not exceed 10 JTU over ambient levels. A secchi disc shall be visible at a minimum depth of 1 meter. All reasonable controls are to be used.	Turbidity shall not exceed 10 JTU over ambient levels. A secchi disc shall be visible at a minimum depth of 1 meter. All reasonable controls are to be used.	Turbidity shall not exceed 25 JTU; a secchi disc shall be visible at a minimum depth of 1 meter.	Turbidity shall not exceed 25 JTU.
Coliform bacteria (per 100 mL)	Not to exceed an arithmetic mean of 20 organisms/100 mL in any group of samples nor shall 10% of the samples exceed 100 organisms/100 mL.	Not to exceed an arithmetic mean of 20 organisms/100 mL in any group of samples nor shall 10 percent of the samples exceed 100 organisms/100 mL.	Not to exceed a log mean of 200 organ- isms/100 mL nor shall 10 percent of the samples exceed 400 organisms/100 mL.	Not to exceed a log mean of 1,000 organ- isms/100 mL nor shall 10 percent of the samples exceed 2,500 organisms/100 mL.
рН	As naturally occurs.	As naturally occurs.	6.5 to 8.0	6.0 to 8.5
Chemical constituents	No chemical constituents in concentrations or combinations which would be harmful to human, animal, or aquatic life for the intended use.	No chemical constituents in concentrations or combinations which would be harmful to human, animal, or aquatic life for the intended use.	No chemical consti- tuents in concentra- tions or combinations which would be harmful to human, animal, or aquatic life for the intended use.	No chemical constituents in concentrations or combinations which would be harmful to human, animal, or aquatic life for the intended use.
(a) Phosphorus	None other than of natural origin.	None other than of natural origin.		
(b) Sodium	Not to exceed 20 mg/L.			

### West Branch Saugatuck River Near Weston, Connecticut

The data from this site indicate water of very good quality (table 11). The measured parameters, with the exception of fecal coliform bacteria, were within the established State standards for class A.

Alkalinity values were lower than the minimum recommended values (table 3) in all samples analyzed. The low alkalinity is believed to occur naturally, because of the absence of carbonate minerals in the soils and bedrock of the drainage area. The pH values measured do not indicate that the low alkalinity poses any problem.

In all samples, fecal coliform bacteria exceeded the criteria. The ratio, (FC/FS) of fecal coliform (FC) to fecal streptococci (FS) bacteria colonies ranged from 0.33 to 0.56 at this site. When the ratio FC/FS is less than or equal to 0.7, and the samples were collected not more than 24 hours flow downstream from the source of pollution, it may be taken as strong evidence that the pollution derives predominately or entirely from nonhuman sources such as livestock or poultry wastes (Millipore, 1973). Wild animals such as ducks, geese, and deer would also be included in this category. Additional sampling would be necessary to determine the origin of the bacteria found at this site.

### Site 2

### Norwalk River at Georgetown, Connecticut

Data indicate the Norwalk River at this site to be of fair quality (tables 6 and 11) and has been rated class C by the State (Connecticut Department of Environmental Protection, 1977). The only parameters exceeding the State standards were pH, which was only slightly elevated on one occasion, and fecal coliform bacteria. The fecal coliform count exceeded 2,500 col/100 mL on several samples, but the log mean value of all samples collected was 522 col/100 mL, which is below the State class C standard (Connecticut Department of Environmental Protection, 1977)

Other physical properties and chemical constituents which were above the criteria of table 3 were color, sodium, iron, lead, manganese, methylene blue active substance (MBAS), and fecal coliform bacteria.

The most consistently high values were found for the total and fecal coliform bacteria. The FC/FS ratios associated with these bacteria counts ranged from 0.36 to 283, indicating that in many instances the bacterial contamination was possibly of human origin. A FC/FS ratio of 4 or greater is considered strong evidence that the bacterial pollution derives from human wastes, if the sample was collected not more than 24 hours flow downstream from the source of the pollution (Millipore, 1973). The source of these bacteria could be sewage

Table 6.--Sumary of water-quality data from site 2, Norwalk River at Georgetown, Conn.

Chemical constituent	Number of		Minimum	Maximum
or physical property	samples	Mean	value	value
or physical property	Samples	Mean	value	value
Streamflow, instantaneous (ft <sup>3</sup> /s)	22	39	1.6	228
Specific conductance (umho/cm at 25°C)	22	289	165	490
pH (units)	22	7.9	7.3	8.6
Temperature, air (°C)	22	14.7	0.0	30.0
Temperature, water (°C)	22	12.1	0.0	22.0
Color (platinum-cobalt units)	19	24	3	65
Turbidity (Jackson turbidity units)	20	3	1	7
	21	11.2	8.6	14.8
Oxygen, dissolved (mg/L)	21	103	93	111
Oxygen, dissolved (percent saturation)	19			20,000
Coliform, total, (col./ 100 mL)		4,100	220	
Coliform, fecal, (col./100 mL)	22	1,000	56	4,500
Streptococci, fecal (col./100 mL)	22	480	3	3,300
Hardness (mg/L as CaCO <sub>3</sub> )	22 22	110 44	62 13	190 110
Hardness, noncarbonate (mg/L CaCO <sub>3</sub> )				
Calcium, dissolved (mg/L as Ca)	22	31	17	61
Magnesium, dissolved (mg/L as Mg)	22	7.8	4.7	12
Sodium, dissolved (mg/L as Na)	22	13	6.9	22
Sodium adsorption ratio	22	0.6	0.4	0.8
Bicarbonate (mg/L as HCO3)	22	79	36	120
Carbonate (mg/L as CO3)	22	0	0	0
Alkalinity (mg/L as CaCO3)	22	65	30	98
Carbon dioxide, dissolved (mg/L as CO <sub>2</sub> )	22	2.1	0.3	5.1
Sulfate, dissolved (mg/L as SO <sub>4</sub> )	19	27	14	43
Chloride, dissolved (mg/L as Cl)	22	33	13	69
Silica, dissolved (mg/L as SiO <sub>2</sub> )	19	6.7	3.3	10
Solids, residue at 180°C,	22	184	77	335
dissolved (mg/L)				
Solids, sum of constituents,	19	163	92	259
dissolved (mg/L)				
Solids, dissolved (tons/acre-ft)	22	0.25	0.10	0.46
Solids, dissolved (tons/day)	22	13.4	1.23	62.8
Solids, residue at 105°C,				
total (mg/L)	22	198	62	382
Nitrogen, NO <sub>2</sub> + NO <sub>3</sub> total (mg/L as N)	22	0.40	0.16	1.0
Nitrogen, ammonia dissolved (mg/L as N)	21	0.10	0.00	0.41
Nitrogen, ammonia dissolved (mg/L as NH2	1) 21	0.13	0.00	0.53
Nitrogen, organic total (mg/L as N)	21	0.44	0.17	1.1
Nitrogen, ammonia + organic total				
(mg/L as N)	22	0.53	0.22	1.3
Nitrogen, total (mg/L as N)	22	0.93	0.40	2.1
Nitrogen, total (mg/L as NO3)	22	4.2	1.8	9.3
Phosphorus, total (mg/L as P)	22	0.11	0.04	0.23
Copper, dissolved (ug/L as Cu)	22	3	0	11
Iron, dissolved (ug/L as Fe)	22	180	40	390
Lead, dissolved (ug/L as Pb)	21	21	2	77
Manganese, dissolved (ug/L as Mn)	21	41	20	90
Zinc, dissolved (ug/L as Zn)	22	40	20	160
Carbon, organic total (mg/L as C)	22	6.4	3.1	13
Methylene blue active substance (mg/L)	16	0.08	0.0	0.90
Sediment, suspended (mg/L)	22	23	3	336
Sediment discharge, suspended (tons/day)		2.2	0.04	32
	,		••••	- <del>-</del>

treatment plant effluents, leachate from septic systems, and/or untreated drains or outfalls. The above normal concentrations of sodium, and the presence of MBAS in some of the samples, along with the concentrations of total phosphorus, total organic carbon, and chloride, also indicate the influence of sewage treatment plants and/or septic systems on the water quality at this site.

The relatively high concentrations of iron, lead, and manganese found at this site probably result from industrial waste discharges.

### Site 3

### Norwalk River at Cannondale, Connecticut

In general, the water quality at this site was assessed as fair (tables 7 and 11), corresponding to its State rating of class C (Connecticut Department of Environmental Protection, 1977). The pH values exceeded the standard on two occasions and were slightly higher than those found upstream at site 2.

Fecal coliform bacteria counts exceeded the criteria in table 3 on many occasions. The FC/FS ratios frequently exceeded 4, indicative of human waste contamination. The source of the bacteria and the above normal sodium concentrations is probably sewage treatment plants and/or leachate from septic systems or outfalls, as were indicated at site 2. The levels of chloride, total phosphorus, and total organic carbon also suggest this type of source. Fecal coliform counts were similar but usually slightly lower here than at site 2.

Iron concentrations exceeded the criteria in three samples, and lead exceeded the criteria in one sample. In general, the concentrations of all metals at this site were lower than upstream at site 2, indicating that no significant concentrations of metals are entering the river between these two sites.

### Site 4

### Comstock Brook at North Wilton, Connecticut

The water quality at this site was excellent (table 11) and has been classified by the State as class AA (Connecticut Department of Environmental Protection, 1977). The analytical results showed only one elevated iron concentration, and several relatively high coliform bacteria counts.

The source of the iron is uncertain, but it may be of natural origin. A large number of the total coliform bacteria do not appear to be of animal origin, based on comparison with the considerably lower fecal coliform and fecal streptococci counts. The origin of these fecal bacteria present cannot be determined by FC/FS ratios, but it is unlikely that the source is human.

Table 7.--Summary of water-quality data from site 3, Norwalk River at Cannondale, Conn.

Chemical constituent	Number of	-	Minimum	Maximum
or physical property	samples	Mean	value	value
Streamflow, instantaneous (ft <sup>3</sup> /s)	22	41	1.7	239
Specific conductance (umho/cm at 25°C)	22	281	165.	440
pH (units)	22	7.9	7.2	8.9
Temperature, air (°C)	22	15.9	1.0	30.0
Temperature, water (°C)	22	12.4	1.0	21.5
Color (platinum-cobalt units)	19	17	0.0	43
Turbidity (Jackson turbidity units)	20	3	1.	7
Oxygen, dissolved (mg/L)	21	11.8	8.2	15.4
Oxygen, dissolved (percent saturation)	21	107	92	136
Coliform, total, (col./ 100 mL)	19	5,800	88	44,000
Coliform, fecal, (col./100 mL)	22	890	19	3,600
Streptococci, fecal (col./100 mL)	22	3,600	13	66,000
Hardness (mg/L as CaCO3)	22	100	62	<b>170</b>
Hardness, noncarbonate (mg/L CaCO3)	22	43	13	87
Calcium, dissolved (mg/L as Ca)	22	29	17	51
Magnesium, dissolved (mg/L as Mg)	22	7.5	4.8	11
Sodium, dissolved (mg/L as Na)	22	13	4.0	23
Sodium adsorption ratio	22	0.6	0.2	0.8
Bicarbonate (mg/L as HCO3)	22	72	43	100
Carbonate (mg/L as CO <sub>3</sub> )	22	0	Ô	0
Alkalinity (mg/L as CaCO <sub>3</sub> )	22	59	35	82
Carbon dioxide, dissolved (mg/L as CO <sub>2</sub> )	22	1.8	0.1	5.1
	19	26	14	41
Sulfate, dissolved (mg/L as SO <sub>4</sub> )	22	32	13	61
Chloride, dissolved (mg/L as Cl)	19	6.5	1.3	10
Silica, dissolved (mg/L as SiO <sub>2</sub> )				
Solids, residue at 180°C,	22	176	101	266
dissolved (mg/L)	10	154	0.2	000
Solids, sum of constituents,	19	154	93	228
dissolved (mg/L)	00	0.04	0.14	0.26
Solids, dissolved (tons/acre-ft)	22	0.24	0.14	0.36
Solids, dissolved (tons/day)	22	14.1	1.20	65.2
Solids, residue at 105°C,	22	192	121	321
total (mg/L)			0.15	
Nitrogen, NO <sub>2</sub> + NO <sub>3</sub> total (mg/L as N)	22	0.41	0.15	1.0
Nitrogen, ammonia dissolved (mg/L as N)	21	0.06	0.01	0.27
Nitrogen, ammonia dissolved (mg/L as NH4		0.07	0.01	0.35
Nitrogen, organic total (mg/L as N)	21	0.47	0.14	1.6
Nitrogen, ammonia + organic total	22	0.53	0.18	1.9
(mg/L as N)				
Nitrogen, total (mg/L as N)	22	0.94	0.47	2.9
Nitrogen, total (mg/L as NO <sub>3</sub> )	22	4.2	2.1	13
Phosphorus, total (mg/L as P)	22	0.11	0.03	0.31
Copper, dissolved (ug/L as Cu)	22	2	0	10
Iron, dissolved (ug/L as Fe)	22	160	10	370
Lead, dissolved (ug/L as Pb)	21	18	4	59
Manganese, dissolved (ug/L as Mn)	21	28	0	50
Zinc, dissolved (ug/L as Zn)	22	40	10	140
Carbon, organic total (mg/L as C)	22	6.1	3.3	9.3
Methylene blue active substance (mg/L)	16	0.01	0.00	0.10
Sediment, suspended (mg/L)	22	21	1	162
Sediment discharge, suspended (tons/day)	22	1.7	0.02	13
		- • •		- <del>-</del>

### Bryant Brook at North Wilton, Connecticut

The data collected at this site indicated excellent water quality (table 11). The State has not specifically listed this stream in its publication (Connecticut Department of Environmental Protection, 1977), but assesses a classification of A to all nontidal waters not listed. The data collected met class A standards with the exception of fecal coliform bacteria.

One sample exceeded the criteria of table 3 for fecal coliform bacteria. The origin of the fecal bacteria cannot be determined by FC/FS ratios. Comparisons of the total coliform counts with fecal coliform counts show that the majority of the coliform bacteria found at this site do not originate from animal waste.

On one occasion, the alkalinity value at this site was found to be lower than the criteria, however, it would require further investigation to determine whether this is a natural occurrence and the frequency of this situation. The pH values, which were above 7.0, indicate that it probably does not have a significant effect on the water quality.

### Site 6

### Norwalk River at South Wilton, Connecticut

The Norwalk River at this site was found to have fair water quality (tables 8 and 11) and has been rated class C by the State (Connecticut Department of Environmental Protection, 1977). Fecal coliform values were frequently above the criteria of table 3. The FC/FS ratios ranged from 0.26 to 21, indicating that on some occasions, the bacteria were probably derived from human sources, while on other occasions, they were not. The fecal coliform counts were usually considerably lower than the total coliform counts, indicating that a substantial portion of the coliform bacteria are not derived from animal wastes. The fact that the fecal coliform bacteria counts at this site are usually less than those found at site 3 suggests that no substantial quantities of animal wastes are entering the river between these sites.

Lead concentrations on several occasions were found to be higher at this site than those found upriver at site 3. The reasons for this, and the source of the lead is not known; however, it appears that additional lead is entering the river between these sites.

For the most part, however, the quality of water at this site is better than that found at site 3. The data show a continuation of the trend seen between sites 2 and 3. with a decrease in the concentrations of most constituents.

Table 8.--Summary of water-quality data from site 6, Norwalk River at South Wilton, Conn.

Chemical constituent	Number	of	Minimum	Maximum
or physical property	sample		value	value
	<del></del>			
Streamflow, instantaneous (ft <sup>3</sup> /s)	22	68	3.8	262
Specific conductance (umho/cm at 25°C)	22	230	134	380
pH (units)	22	7.7	7.1	8.5
Temperature, air (°C)	22	16.6	0.5	28.5
Temperature, water (°C)	22	12.3	1.0	22.0
Color (platinum-cobalt units)	19	15	. 3	38
Turbidity (Jackson turbidity units)	20	2	1	6
Oxygen, dissolved (mg/L)	22	11.8	9.0	15.6
Oxygen, dissolved (percent saturation)	22	107	97	119
Coliform, total, (col./ 100 mL)	21	10,000	20	66,000
Coliform, fecal, (col./100 mL)	22	410	14	3,100
Streptococci, fecal (col./100 mL)	22	420	6	3,300
Hardness (mg/L as CaCO3)	22	75	23	98
Hardness, noncarbonate (mg/L CaCO <sub>3</sub> )	22	32	10	55
Calcium, dissolved (mg/L as Ca)	22	21	5.8	28
Magnesium, dissolved (mg/L as Mg)	22	5.6	1.8	7.5
Sodium, dissolved (mg/L as Na)	22	13	7.1	25
Sodium adsorption ratio	22	0.7	0.4	1.1
Bicarbonate (mg/L as HCO3)	22	52	16	69
Carbonate (mg/L as CO <sub>3</sub> )	22	0	0	0
Alkalinity (mg/L as CaCO <sub>3</sub> )	22	43	13	57
Carbon dioxide, dissolved (mg/L as CO <sub>2</sub> )	22	2.2	0.2	5.2
Sulfate, dissolved (mg/L as SO4)	19	21	7.6	33
Chloride, dissolved (mg/L as Cl)	22	28	8.8	58
Silica, dissolved (mg/L as SiO <sub>2</sub> )	19	7.2	4.0	10
Solids, residue at 180°C,	22	139	59	208
dissolved (mg/L)	10	100	47	101
Solids, sum of constituents,	19	122	47	181
dissolved (mg/L)	22	0.19	0.08	0.28
Solids, dissolved (tons/acre-ft)	22	20 1	2.12	65.1
Solids, dissolved (tons/day)	22	155	55	237
Solids, residue at 105°C, total (mg/L)	44	155	33	237
Nitrogen, NO <sub>2</sub> + NO <sub>3</sub> total (mg/L as N)	22	0.43	0.18	0.94
Nitrogen, ammonia dissolved (mg/L as N)	22	0.04	0.00	0.10
Nitrogen, ammonia dissolved (mg/L as NH <sub>4</sub>		0.05	0.00	
Nitrogen, organic total (mg/L as N)	22	0.41	0.07	1.3
Nitrogen, ammonia + organic total	22	0.42	0.10	1.3
(mg/L as N)		0.12	0.10	2.0
Nitrogen, total (mg/L as N)	22	0.86	0.31	1.8
Nitrogen, total (mg/L as NO <sub>3</sub> )	22	3.8	1.4	7.9
Phosphorus, total (mg/L as P)	22	0.05	0.01	0.13
Copper, dissolved (ug/L as Cu)	22	3	0	10
Iron, dissolved (ug/L as Fe)	22	150	30	270
Lead, dissolved (ug/L as Pb)	22	20	2	85
Manganese, dissolved (ug/L as Mn)	22	26	Õ	50
Zinc, dissolved (ug/L as Zn)	22	20	ŏ	60
Carbon, organic total (mg/L as C)	22	5.5	2.4	11
Methylene blue active substance (mg/L)	16	0.01	0.00	0.10
Sediment, suspended (mg/L)	22	30	1.0	389
Sediment discharge, suspended (tons/day)		5.1	0.04	69
			-	

### East Branch Silvermine River near Ridgefield, Connecticut

This reach of river is rated class AA by the State (Connecticut Department of Environmental Protection, 1977), and the data collected indicated relatively good water quality (table 11).

In one sample, the fecal coliform bacteria count exceeded the State Class AA standard. However, the other fecal coliform counts were well below the standard, and fecal streptococci counts were relatively low. The FC/FS ratios associated with these counts tend to indicate a non-human source.

Of note at this site were the relatively high concentrations of manganese found in all three samples collected. Further investigation would be necessary to determine the exact source, but it is possible that it occurs naturally as a result of the mineral content of the rocks and soils of the drainage basin.

Other values found were within the suggested and established criteria and were not indicative of any significant contamination.

### Site 8

### East Branch Silvermine River near North Wilton, Connecticut

The State has rated this reach of the river as class AA (Connecticut Department of Environmental Protection, 1977), and the data collected indicate very good water quality (table 11). Although one total coliform count was relatively high, the fecal coliform and fecal streptococci counts were generally low, indicating that at the time of sampling, fecal waste contamination was insignificant.

Values for nearly all parameters measured at this site showed an improvement in water quality over that found at site 7. A possible exception was MBAS, which was detected in two out of the three samples collected at this site, and was not found in any samples collected at site 7. Although the concentrations of MBAS were well below the water-quality criteria (table 3), its presence could be significant, representing natural organic acid complexes or some type of foaming agents that could be entering the river between this site and site 7.

### Thayers Brook near Silvermine, Connecticut

This stream has not been specifically classified by the State (Connecticut Department of Environmental Protection, 1977), but was assumed to be class A, as was the case with Bryant Brook (site 5). The data collected indicated water of good quality (table 11).

Total coliform bacteria counts ranged from 50 to 1400 col/100 mL. The fecal coliform counts were considerably lower by comparison, indicating that the majority of the coliform bacteria did not come from fecal material. In two of three samples, fecal coliform counts exceeded 20, however, the FC/FS ratio associated with the highest bacteria counts measured (May 12, 1976) was 0.51, indicating that the probable sources of these fecal bacteria were some warm-blooded animals other than humans. Because the numbers of these bacteria were relatively low, it is possible that they originated from the natural populations of birds and mammals in the area.

The pH values at this site were consistently lower than those found at any other site in the study area. The samples on all three occasions also had low alkalinity values. These low values may be partly due to the geologic materials in the basin that affect the chemical characteristics of ground-water contributions to streamflow, and/or organic decomposition and other biological activity occurring in the pond just upstream of the sample site (fig. 1).

Some iron and manganese concentrations were also above the criteria. The source of these constituents is also probably related to the geologic material of the basin, and/or biological activity in the pond.

### Site 10

### Silvermine River near Silvermine, Connecticut

The State has classified this reach of the river as class B (Connecticut Department of Environmental Protection, 1977), and the data collected in this investigation are indicative of good water quality (table 11).

Values for most parameters analyzed at this site showed an improvement in the quality of the water from that which was found in the East Branch of the river at sites 7 and 8.

One relatively high total coliform count was found (2000 col/100 mL), with a corresponding fecal coliform count of 330 col/100 mL. The FC/FS ratio for this sample (May 12, 1976) was 0.69, indicating that the source of the fecal bacteria was probably not human waste.

Slightly below criteria alkalinity values were found at this site on two of the three samples collected, but it does not appear to be a significant factor in the overall quality of the stream, based on pH values.

### **Bottom Materials**

Bottom materials have a significant role in the water quality of a stream for several reasons. Of primary importance is the fact that many chemical constituents of water can be sorbed onto these bottom sediments, and in effect, they act as a sink for them. These chemicals can then be transported with the sediments either by chemical, physical, and biological processes. It is also possible for some of these chemicals to be brought back into solution. Thus, bottom materials can be thought of as an indicator of the chemicals that have been in the water, as a transport mechanism for the chemicals, and as a potential source of these chemicals at a future time.

Bottom material samples were collected at sites 2, 3, and 4 in June of 1976 (table 9). Values from Comstock Brook at site 4 were used to reflect natural conditions in the area based on land use in its hasin; therefore, the levels of aluminum, copper, and iron appear to be elevated at Norwalk River sites 2 and 3. Comparison of data between these two sites indicates that metal concentrations are higher at site 3 than at site 2.

Because of the mobility of bottom materials, and the lack of information concerning their rate of transport in the river, it is not possible to determine where the metals originated in the basin. More detailed sampling over an extended period of time would be necessary to determine whether these materials are actually entering the river between sites 2 and 3.

Low concentrations of the pesticides DDE and DDT, and a rather high concentration of chlordane (45 ug/kg) were also found in the sample from Comstock Brook (site 4), indicating that these substances were used in the basin at some time in the past.

### <u>Biological Studies</u>

The diversity of the biological community found in a body of water is a good indication of the relative water quality. In general, water of good quality will support a wide variety of organisms, with relatively small numbers of each type of organism present. On the other hand, water of poor quality usually supports a rather sparse variety of organisms, with those that are tolerant of the poor quality being present in large numbers.

Data for biological analysis were collected at sites 2, 3, and 4 from October 1975 to June 1976 (table 10).

Samples of phytoplankton populations in the Norwalk River (sites 2 and 3) showed good diversity, indicative of rather good water quality. There was less diversity of phytoplankton in Comstock Brook (site 4); however, the population was made up entirely of various species of diatoms. Diatoms are usually indicative of good water quality, and the lack of green or blue-green algae would tend to indicate a deficiency of some plant nutrient as the limiting factor for diversity at this site. Data for algal growth potential and total phytoplankton cells found at sites 2 and 3 were considerably higher that at site 4.

Benthic invertebrates possess several characteristics that make them valuable indicators of water quality. Because of their limited mobility and relatively long life span, populations of these bottom dwelling animals reflect water-quality conditions over a long period of time, not just at the instant of sample collection. Information is also available concerning the specific water-quality conditions under which various types of these organisms can exist. Thus, the absence or presence of certain types of benthic invertebrates is indicative of specific water-quality conditions.

Benthic invertebrate populations in the Norwalk River at sites 2 and 3 showed relatively good diversity, which is indicative of good quality. However, the predominant types of organisms present would be characterized as those which are relatively tolerant to pollutants, such as midges and caddis flies. The diversity of benthic invertebrates at site 4 in Comstock Brook was less than that at the Norwalk River sites, possibly due to a limited amount of available food or habitat. The chlordane concentrations found in the bed sample at this site could also be indicative of a condition that existed in the past, which may explain the lesser diversity at this site. Considerably more data would be necessary to determine if this were the case. The types of benthic invertebrates found at this site included several species of mayflies, which are considered to be intolerant of pollution as a group and indicative of good water quality.

### SUMMARY

The data collected indicate that streams in the Town of Wilton, Connecticut had fair to excellent water quality. The parameters which most often exceeded recommended and established criteria (table 3) were fecal coliform bacteria, pH, alkalinity, manganese, and iron.

The total coliform bacteria population in most instances appear to be composed primarily of types that are not derived from fecal material. When fecal coliform bacteria counts were found to exceed the criteria, the origins of these bacteria, based on FC/FS ratios, were usually not related to human waste. The exceptions are sites 2, 3, and 6 on the Norwalk River, which had FC/FS ratios indicative of human waste contamination on several occasions.

The Norwalk River showed improvement in water quality as it flowed through Wilton. The concentrations of most chemical constituents and bacteria decreased from the Georgetown site to the Cannondale site. Data from the South Wilton site showed a continuation of this trend of improved quality, with some exceptions, most notably lead, which appears to be entering the river between the Cannondale and South Wilton sites.

In general, the East Branch Silvermine River had better water quality than that of the Norwalk River. The data showed that the quality of the Silvermine River also improved as it flowed through Wilton. Nearly all parameters showed decreases in concentration between the Ridgefield and the North Wilton sites, with the exception of the methylene blue active substance. This substance was present in low concentration at the North Wilton site, and absent upstream at Ridgefield. Data from further downriver at Silvermine, on the main stem of the Silvermine River, showed more improvement in water quality.

Bottom material analysis from the Norwalk River (sites 2 and 3) and Comstock Brook (site 4) showed concentrations of some metals in the Norwalk River above those expected under normal conditions. The Comstock Brook sample had no abnormal concentrations of metals, but did contain a high concentration of chlordane, and trace concentrations of DDE and DDT.

Biological samples collected at sites 2, 3, and 4 indicated relatively good water quality. In general, the types of organisms found in the Norwalk River were indicative of higher concentrations of nutrients, and were of types that are more tolerant to degraded water quality than those found in Comstock Brook.

<u>Table 9.--Bottom material analyses</u>
Site 2, Norwalk River at Georgetown, Conn.

DATE	ALUM- INUM. RECOV. FM BOT- TOM MA- TERIAL (UG/G)	BARIO RECO FM BO TOM I TER (UG,	OV. LI OT- RE MA- FM IAL TOM /G TE	UM. R COV. FM BOT- TO MA- T HIAL (		CHRO- MIUM. RECOV. M BOT- OM MA- TERIAL (UG/G)	COBALT. RECOV. FM BOT- TOM MA- TERIAL (UG/G AS CO)	COPPER• RECOV• FM BOT- TOM MA- TERIAL (UG/G AS CU)	IRUN. RECOV. FM BOT- TOM MA- TERIAL (UG/G AS FE)	TERIAL TERIAL TOM MA- TERIAL TOM MA-	FM BOT-
JUN 21	2900		55	0	10	6	4	60	6000	<b>i1</b> 0	360
DATE	MOLYH- DENUM. PECOV. FM BOT- TOM MA- TERIAL (UG/G)	NICKE RECO FM BO TOM F TER: (UG)	DV. REDT- FM MA- TOM IAL TE // (U	COV. BOT- R MA- FM RIAL TO G/G T	RO1- 1	TI- ANIUM, TOTAL N BOT- OM MA- TERIAL (UG/G)	VANA- DIUM. TOTAL IN BOT- TOM MA- TERIAL (UG/G)	PCB, TOTAL IN BOT- TOM MA- TERIAL (UG/KG)	ALDRIN. TOTAL IN BOT- TOM MA- TERIAL (UG/KG)	CHLOR- DANE, TOTAL IN BOT- TOM MA- TERIAL (UG/KG)	DDD, TOTAL IN BOT- TOM MA- TERIAL (UG/KG)
JUN 21	3		12	0	13	90	<b>9.</b> 0	. 0	• 0	0	• 0
71 PQ	TO IN E TOM TEF TE (UG)	MA- 1	DDT. TOTAL IN BOT- FOM MA- TERIAL (UG/KG)	DI- AZINON. TOTAL IN BOT- TOM MA- TERIAL (UG/KG)	DI- ELDRIN TOTAL IN BOT TOM MA TERIA (UG/KG	TOT - IN B - TOM L TER ) (UG/	AL TOT OT- IN 8 MA- TOM IAL TER	ON. CHL AL TOT OT- IN E MA- TOM IAL TER	OR. CHL TAL EPOX BOT- TOT. MA- BOT RIAL MA	IDE TO IN TOM TOM	DANE TAL BOT- MA- RIAL /KG)
D <b>A</b>	TOT IN F TOM TEF	ION+ FAL POT- 1 MA- PIAL	PARA- THION. TOT. IN BOTTOM MATL. (UG/KG)	METHYL TRI- THION, TOT, IN MOTTOM MATL,	PARA- THION TOTAL IN BOT TOM MA TERIA (UG/KG	PHE TOT: IN B TOM:	NE THI AL TOT OT- IN B MA- TOM IAL TER	AL FA OT- UIA MA- % FI IAL TH	NT. MA NLL FA NM. DIA NER % FI NAN TH	T. M. LL F. M. DI. NER % F. IAN TI	E1) AT. ALL AM. INER HAN 0 MM
NUL 1S	•••	• 0	•0	•0	•	0	0	•0	1	3	100

Table 9.--Bottom material analyses.--Continued

				Table	91	Bottom	mater	ial a	Table 9Bottom material analysesContinued	Con	tinned				
				Site		lorwalk	Rive	rato	3, Norwalk River at Cannondale,		Conn.				
DATE	ALUM- INUM. RECOV. FM BOT- TOM MA- TERIAL		BARIUM. RECOV. FM BOT- TOM MA- TERIAL (UG/G	BERYL- LIUM. RECOV. FM BOT- TOM MA- TERIAL	<u>u.</u> ⊷	BORON. RECOV. FM BUT- TOM MA- TERIAL (UG/G	CHE MILL TOM E	CHRO- MIUM. RECOV. FM BOT- TOM MA- TEKIAL	ECOBALT. RECOV. FM BOT- TOM MA- TERIAL (UG/G	3 LF		IRON• RECOV• FM BOT- TOM MA- TERIAL (UG/G AS FE)		LEAU. RECOV. FM BOT- TOM MA- TERIAL (UG/G	MANGA- NESE• RECOV• FM BOT- TOM MA- TERIAL
JUN 21	4200	00	95		~	10		Þ	7		190	8900		150	094
NATE	MOLYB- DENUM. RECOV. FM BOT- TOM MA- TERIAL	2 4 -	NICKEL. RECOV. FM BOT- TOM MA- TERIAL (UG/G	SILVER. RECOV. FM BOT- TOM MA- TERIAL (UG/G	<b>u</b> –	STRON- TIUM. RECOV. FM BOT- TOM MA- TERIAL		TI- TANIUM. TOTAL IN BOT- TOM MA- TERIAL (UG/G)	VANA- DIUM• TOTAL IN BOT- TOM MA- TERIAL	PCB. TOTAL IN BOT- TOM MA- TERIAL (UG/KG)		ALDMIN. TOTAL IN BOT- TOM MA- TERIAL	H ~	CHLOM- DANE • TOTAL IN BOT- TOM MA- TERIAL	DDD. TUTAL IN HOT- TOM MA- TEMIAL (UG/KG)
JUN 21		4	91		0	20		170	13		0	0•		0	0.
0.04	T IN TO DATE (U	DDE. TOTAL IN HOT- TOM MA- TERIAL	UDT. TOTAL IN BOT- TOM MA- TERIAL (UG/KG)	- •	DI- AZINON• TOTAL IN HOT- TOM MA- TERIAL (U6/KG)		DI- ELDRIN. TOTAL IN BOI- TOM MA- TERIAL (UG/KG)	ENDRIN. TOTAL IN BOT- TOM MA- TERIAL (UG/KG)		ETHION. TOTAL IN BOT- TOM MA- TERIAL (UG/KG)	HEPTA- CHLOM, 101AL IN BOT- TUM MA- TEPIAL	2.	HEPTA- CHLOR EPOXIUE TOT. IN BOTTOM MATL.	LINDANE TOTAL IN BOT- TOM MA- TERIAL	9 1-446 1-466
JUN 21.5	UN 21	•		0	•	0•	0		0	0,	•	0	0•		0.
J.	T TO	MALA- THION. TOTAL IN GOT- TOM MA- TFRIAL	METHYL PARA- THION. TOT. IN BOTTOM MATL. (UG/KG)	• 2 5 • ~	METHYL TRI- THION. TOT. IN BOTTUM MATL.		PARA- THION. TOTAL IN BOT- TOM MA- TERIAL	TOXA-PHENE, TOTAL IN BOT-TOM MA-TERIAL (UG/KG)	·	TRI- THION. TOTAL IN BOT- TOM MA- TERIAL	BED MAT. FALL UIAM. % FINER THAN	·	HED MAT. FALL DIAM. * FINER THAN	BEU MAT. FALL UIAM. * FIVER THAN	α <u>Σ</u>
JUN 21.	UN 21	0•		0•	•	0	0.		0	9		-	8	1	100

Table 9.--Bottom material analyses.--Continued Site 4, Comstock Brook at North Wilton, Conn.

-	DATE	JUN 22	DATE	JUN 22	DATE	JUN 22	DATE	JUN 22.
ALUM- INUM• RECOV• FM BOT-	TERIAL (UG/G)	2100	NICKEL. RECOV. FM BOT. TOM MA- TERIAL (UG/G	:	DDE. TOTAL IN BOT- TOM MA- TERIAL (UG/KG)	:	MALA- THION. TOTAL IN BOT- TOM MA- TERIAL	:
BARIUM. RECOV. FM BOT-	16 K 1 A L (UG/G AS BA)	0 †		<b>,</b>	•	•	Arton 3	0
LL F		•	SILVER. RECOV. FM BOT- TOM MA- TERIAL (UG/G AS AG)	0	DDT. TOTAL IN BOT- TOM MA- TERIAL (UG/KG)	7.7	METHYL PARA- THION. TOI. IN BOTTOM MATL.	0.
BERYL- CLIUM. RECOV. HEECOV.		0	STRON- TIUM. RECOV. FM HOT- TOM MA- TERIAL	7 0	DI- AZINON, TUTAL IN BOT- TOM MA- TERIAL (UG/KG)	•	METHYL TRI- THION. TOI. IN BOTTOM :MATL.	•
	TERIAL (UG/G)	7	11- TANIUM. 101AL. IN BOT- TOM MA- TERIAL		DI- ELDRIN. TOTAL IN BOT- TOM MA- TERIAL	0	PARA- THION- TOTAL N IN BOT- TOM MA- TERIAL	2
COBALT. RECOV. FM BOT- TOM MA-	(UG/G AS CO)	e.		130		0.	•	<b>5</b>
0 14-	(UG/G AS CU)	_	VANA- DIUM. TOTAL IN BOT- TOM MA- TERIAL	0.9	ENDRIN. TOTAL IN BOT- TOM MA- TERIAL (UG/KG)	0.	TOXA- PHENE + TOTAL IN BOT- TOM MA- TERIAL	0
<b>Ŀ</b> ⊢		•	PCB. TOTAL IN BOT- TOM MA- TERIAL (UG/KG)	Э	ETHION. TOTAL TO BOT- TOM MA- TERIAL	9.	TRI- THION. TOTAL IN BOI- TOM MA- TERIAL	o.
u	16K1AL (UG/G (1	2900	ALDPIN. TUTAL IN BOT- TOM MA- TERIAL (UG/KG)	•	HEPTA- CHLOR. TOTAL IN BOT- TOM MA- TEMIAL	0	BEU MAT. FALL DIAM. % FINER THAN	~
	16 A I A I A I A I A I A I A I A I A I A	130	CHLOR- DANE. TOTAL IN BOT- TOM MA- TERIAL	7	HEPTA- CHLOR EPUXIDE TOT. IN BOTTOM MATL.	•	bED MAT. FALL DIAM. % FINER THAN	
MANGA- NESE+ RECOV- FM BOT-	TERIAL (UG/G)	300	DDD. 101AL 10 BOT- 10M MA- 1ERIAL	10	LINDANE TOTAL 10 BOT- TOM MA- TERIAL	0	BED MAT. FALL UIA". ₹ FIVER 7 ¥ FIVER	S
MOLYB- DENUM. RECOV. FM BOT-	TERIAL (UG/G)	4	APL 1141 701 701	0	A A A A A A A A A A A A A A A A A A A	0	A Z E	100

### Table 10.--Biological analyses

Site 2, Norwalk River at Georgetown, Conn.

	ORGANISM.	101	••••HYDROPSYCHE
OFGANISMS EQUAL TO OR GREATER THAN 158	DOMINANT OFGANISMS	464	CHEUMATOPSYCHE
			• • • I YUROPSYCHIUAE
17 1	•••• VOX # I TOCO P	F	TOTOLOG TEDA
	••• 50% FX CC BC	-	• • • • • • • • • • • • • • • • • • •
33 2	· · · · NITESCHIA		• PLECOPTERA
	NITZSCHIACE AE	<b>p</b>	BOYERIA
300# 22	· · · NAVICUCA	•	• • • • • • • • • • • • • • • • • • •
	NAVICULACEAE		ODONATA
* 0	• • • • • • • • • • • • • • • • • • •	Lo	BAETIS
	MEXICIONACE AE	ř	BAETIDAE
17 1	SYNEDRA		. EPHEMEROPTERA
350# 26	FRAGILARIA	•	ANTOCHA
	FRAGILARIACEAE		TIPULIDAE
17 1	CYMBELLA	821	SIMULIUM
	CYMBELLACEAE	•	SIMULIIDAE
* (	RHOICOSPHENIA	-	HEMERODROMIA
1/ 1	COCCONE IS	•	EMPIUIDAE
150 4	ACHNANTHES	270	THIENEMANNIELLA
	ACHNANTHACE AF	140	RHEOTANY TARSUS
	. PENNALE'S	04	POLYPEDILUM
35 2	MELOSIRA	P.O.	PENTANEURA
350# 26	CYCLOTELLA	460	EUKIEFFERIELLA
	COSCINODISCACEAE	210	CRICOTOPUS
	CENTRALES	30	CORYNONE UP A
	. HACILLARIOPHYCEAE	70	• • • CARDIOCLADIUS
	CHRYSOPHYTA	60	BRILLIA
			CHIRONOMIDAE
61 5	CHLAMYDUMONAS		UIPTERA
	CHLAMYDOMONADACEAE	_	ECTOPHIA
	VOL VOCALES		• • • PSEPHENIDAE
17 1	SCENEDESMUS		COLEOPTERA
	SCENEDE SMACE AE		INSECTA
1/ 1	SCHROEDERIA	20	UNKNOWN GENUS
	CHARACIACEAE		UNKNOWN FAMILY
	CHLOROCOCCALES		HYDRACARINA
	CHLOROPHYTA (GREEN ALGAE)	•	ARTHROPODA (ARTHROPODS)
746		000	G P P P P P P P P P P P P P P P P P P P
S	ODGANISH	CHAT	
		C	•••• GENOS-1 NSEC   #
C • 1	•••• 65000	א ני	· · · · · · · · · · · · · · · · · · ·
<b>**</b> **********************************	*** AM L T	. 4	** P31C1
- 4	• • CX:: TX		• • ORDER
	•CEA55	•	•CLASS
4.0	DIVERSITY: DIVISION	0.0	DIVERSITY: PHYLUM
1300	TOTAL CELLS/ML	1607	TOTAL COUNT
		i I	
JUN 21.76 1005	DATE	JUN 21.76 1005	UATE Time
BER 1975 TO JUNE 1976	PHYTOPLANKTON ANALYSES, OCTOBER 1975 TO	ER 1475 TO JUNE 1976	BENTHIC INVERTEBRATE ANALYSES. OCTOBER 1975 TO JUNE

### Table 10. -- Biological analyses. -- continued

## Site 3, Norwalk River at Cannondale, Conn.

OCTOBER 1975 TO JUNE 1976	JUN 21,76 1330	096	\$.00 \$.0 \$.1 \$.5	CELLS PER-	17 6	17 5 130 14 • 0	130 14	390# 40 34 4	↑ #3↑ T	ORGANISM: MAY NUI HAVE BEEN COUNTEU: LESS THAN 1/2+	
PHYTOPLANKTON ANALYSES, OCTOBER 1975 TO JUNE	DATE	TOTAL CELLS/ML	DIVERSITY: DIVISION CLASS ORDER THE STATE	ORGANISM	CHLONOPHYTA (GREEN ALGAF) CHLONOPHYCEAE CHUNCOCCALES CHANACIACAE CHANACIACAE CHANACIACAE CONOCYCLAFAE	**************************************	VOLVOCALESVOLVOCACEAEPANJOMINA	CHMYSOPHYTA  JACILLAPIOPHYCEAE  CENTIALES  COSCINODISCACEAE  CYCYCLOTELLA  PENNALES  ACHNANTAGEAE	RHOICOSPHENIA NAVICULACEAE NAVICULA	NOTE: # - DOMINANT ORGAN • - OBSERVED ORGAN	.00
ANALYSES, OCTOBER 1975 TO JUNE 1976	JUN 21.76 1330	3828	0 9 5 5 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	COUNT	v	7	F 6 6 7	945 6 66 994 131 35	~ -	0- 0 - 4 54 0 - 1	-
BENTHIC INVERTEBRATE ANALYSES. OCT	DATE TIME	TOTAL COUNT	DIVERSITY: PHYLUM CLASS CORDER TO SOMBER TO SO	ORGANISM	ARTHROPODA (ARTHROPODS) ARACHNIDA	COLEOPTERA ELMIDAE OPTIOSERVUS	CHIRONOMIDAEBHILLIACORYNDNEURA	EURIEFFERIELLAPUNIEFFERIELLAPUNIEFFERIELLAPUNIEFFERIELLAPUNIEFFERIELLAPUNITARSUSTANTARSUSTHIENEMANNIELLASIMULIUM	IPULIDAE anioCha TIPULA EPHEMEROPTERA	HAETIDAE HAETIDAE HAETIDAE HAETIDAE FSEUDOCLOEON HEVRAGENIIDAE STENONEHA HOVERIA HOVERIA HOVERIA PARAGNETINA PERLIDAE PARAGNETINA PERLIDAE HYDROPSYCHE HYDROPS	PHYSA

### Table 10. -- Biological analyses. -- continued

# Site 4, Comstock Brook at North Wilton, Conn.

ICTOBER 1975 TO JUNE 1976	JU14 22.76 1240	120	カガ マ マ マ マ マ マ マ マ マ マ マ マ マ マ マ マ マ マ マ	r • -	CELLS PLW-			/5 to 5		0	•	9 ~		\$ <b>,</b>			4	o	MASS FOLLOWS TO CALLANA TERMS	- OBSERVED ORGANISM. MAY NOT MAVE BEEN COUNTED! LESS THAN 1/24																
PHYTOPLANKTON ANALYSES. OCTOBER 1975 TO JUNE 1976	DATE	TOTAL CELLS/ML	DIVERSITY: UIVISION CLASSORDER	· · · GENUS	ORGANISM	CHRYSOPHYTA .BACILLARIOPHYCEAE	• • PENNALES • • • • ACMNANTHACEAE	ACHNANTHES	CYMBELL ACEAE	EUNOTIACEAE	EUNOTIA	GOMPHONEMA	NAV I CUL ACEAE	· · · · NAVICULA	EUGLENOPHYTA (EUGLENOIDS)	.EUGLENUPHYCEAE EUGLENALES	EUGLENACEAE	223300000000000000000000000000000000000																		28
ANALYSES. OCTOBER 1975 TO JUNE 1976	JUN 22,76 1240	590	0000 11100	2.5 2.6	COUNT			·			-	•	ć.	10	97	170	10		- <b>-</b>	,	7)	~	~	•	٥٥		0.1		<b>-</b>	~					~	
BENTHIC INVENTEBRATE ANALYSES. OCTO	DATE. TIME	TOTAL COUNT	DIVERSITY: PHYLUM CLASSORDERFAHILY	••••6ENUS ••••6ENUS-INSECTA	ORGAN I SM	ANNEL IDA OL 160CHAETA	PLESIOPORA Tubilicidae	UNKNOWN GENUS	ARTHRUPUDA (ARTHROPODS)	COLEOPTERA	ELMIDAE	DIPTERA	CHIRONOMIDAE	CHYPTOCHIRONOMUS	EUKIEFFERIELLA MICHOTOFO	**************************************	TANYTARSUS	TIPULIDAE	EPHEMEROPTERA	BAETIDAE	••••CAENIDAE	CAENIS	EPHEMERELLIDAE	HEPTAGENIDAE	STENONEMA PLECOPTERA	PERLIDAE	TPICHOPTERA	POLYCENTROPODIDAE	****NYCLIOPINICAX	UNKNOWN GENUS	CNIDARIA (CNIDARIANS)	. HYDROIDA	HYURIDAE	4D1 USCA (#01 USCS)	PHYSIDAE	

Table 11.--Water-quality analyses

Site 1, West Branch Saugatuck River near Weston, Conn.

COLI- FORM. TOTAL. IMMED. (COLS. PER	12	1300	430	CAR- HONATE (MG/L AS CO3)	0	0	0	1+0	50	46	33	- u u u	00	00.	00.
OXYGEN. OIS- SOLVED (PEH- CENT SATUR- ATION)	101	102	\$	BICAR- BONATE (MG/L AS HC03)	16	15	21	NITRO- GEN-AM- MONIA + ORGAVIC TOTAL (MG/L AS V)	":	•	•	METHY- LEVE RL'JE ACTIVE SUN- STANCE MG/L)	•		
5	11.5	0.1	φ. • •		۲.	9.	٠	NITRO- GEN• NOZ+NO3 TOTAL (MG/L AS N)	16.	.19	.39	CARBON. ORGANIC TOTAL (MG/L AS C)	2.4	4.7	0.9
5		2	~		_	•	s	SOLIDS. MESIDUE AT 105 DEG. C. TOTAL (MG/L)	9	63	63	ZINC. DIS- SULVEU (UG/L AS ZN)	Э	0	>
10k 117 ()10)		•••		SODIUM. DIS- SOLVEU (MG/L AS NA)	7.1	, ,	7.6		4.34	<b>2.</b> 6	1.34		20	30	30
TEMPER- ATURE. WATER (DEG C)	12.0	16.0	22.5	MAGNE- SIUM. DIS- SOLVED (MG/L AS MG)	2.1	2.1	2.3	SOL IDS. UIS- SOLVEU (TONS PEH DAY)		-		MANGA- NE SE DIS- SOL VED (UG/L AS MN)			
	3.0	8.5	30.5		0.7	5.9	<b>6.</b> 1	SOLIDS. DIS- SOLVED (TONS PER AC-FT)	•10	20.	.11	IRDN. DIS- SOLVED (UG/L AS FE)	06	150	29 29
<b>–</b>	٧				•			SOL 105. RESIDUE AT 180 DEG. C DIS- SOLVED (MG/L)	7.3	4	52	COPPER. DIS- SOLVED (UG/L AS CU)	0	0	o
PH (UNITS)	1.2	7.1	6.9	HARD- NESS. HONCAR- HONATE (MG/L CACU3)	13	10	12		_			3	.020	050	050
SPE- CIFIC CON- DUCT- ANCE (MICKO- MHOS)	116	95	111	HAKD- NESS (MG/L AS CACO3)	98	23	30	CMLO- RIDE: 01S- SOLVED (MG/L AS CL)	13	01	71	PHOS- PHORUS- TOTAL (MG/L	•	•	3
	~	13	۶.		12	100	230	CARBON DIOXIDE UIS- SOLVED (MG/L AS CO2)	1.6	1.9	4.2	NITRO- GEN• TOTAL (MG/L AS NO3)	2.3	7.6	3.2
STREAM- FLOW. INSTAN- TANEOUS	~			STREP- 10COCCL FECAL- (COLS- PER 100 ML)	4	-	c	ALKA- LINITY (MG/L AS CACO3)	13	12	11	NITRO- GÉN• TOTAL (MG/L AS N)	.51	.59	.72
<b>∃I</b> ≇E	1425	1415	1405	COLI- FORM. FECAL. 0.45 UM-MF (COLS./	•	420	130		:	:	:		:	:	:
DATE	APR 14.	12	23	OATE	APR 14	MAY 12	23	DATE	APR.	MAY 12.	23.	DATE	APR 14	12.	23.

Table 11.--Water-quality analyses.--continued

Site 2, Norwalk River at Georgetown, Conn.--Continued

SODIUM. DIS- SOLVED (MG/L AS NA)	10	4.6	11	50	6.9	13	6.8	12	13	18	22	12	12	12	16	18	50	12	12	2.5	1.1	14
MAGNE- SIUM. DIS- SOLVED (MG/L AS MG)	6.1	4.0	7.7	æ• 6	4.7	4.1	7.1	4.1	8.0	8.2	9.2	₽. 9•6	7.9	1.2	8.7	11	12	7.0	7.4	5.5	₩. 4	7.3
CALCIUM UIS- SOLVED (MG/L AS CA)	25	7.	27	61	1.1	7 7	5.0	53	39	38	47	19	27	54	4	97	75	5.7	55	18	1.1	23
HARU- NESS, NONCAR- BONATE (MG/L CACO3)	27	16	56	110	13	6	20	34	99	53	12	4.1	46	82	в	7.0	95	15	0 1	16	56	97
HARD- NESS (MG/L AS CACO3)	£,	74	56	190	62	150	79	110	130	130	160	7.1	100	06	150	160	150	2	63	9	65	88
STREP- TOCOCCI FECAL, (COLS, PER	0 7	7.0	110	880	3300	620	1	:	1	1	:	:	;	1	;	:	1	:	;	•	;	1
STREP- TOCOCCI FLCAL• KF AGAR (COLS• PEH	;	:	ł	ł	ł	1	120	45	<b>J</b>	æ	e	38	•	81	130	230	850	1600	190	2200	30	84
COLI- FORM, FECAL, UM-MF (COLS,/	:	:	:	:	1	1	170	100	230	56	540	160	1700	550	1000	099	1300	4500	700	3100	160	360
COLI- FORM. FECAL. 0.45 UM-MF (COLS./	110	390	1800	290	1200	2900	;	:	:	;	!	;	:	;	;	1	:	1	:	1	:	1
COLI- FORM. TOTAL. IMMEU. (COLS. PER	1976 700	1700	6500	3800	;	10000	1400	006	650	220	1000	320	1	3900	3400	640	20000	0009	7200	7600		1600
DATE	APR . 1	10	200 21	JU. 13.	11	21	27	22.	29	26	23	29	100.	111	22.	25	23	21	19.	•	22	APR . 19

Table 11.--Water-quality analyses.--continued Site 2, Norwalk River at Georgetown, Conn.

OXYGEN. DIS- SOLVEU (PER- CENT SATUR- ATION)	104	106	96	93	D D	16	102	102	;	101	101	107	106	104	110	111	ድ	101	102	111	104	101
OXYGFN, DIS- SOLVED (MG/L)	12.6	11.1	8.6	9.0	60	0.6	12.5	13.7	;	14.8	14.4	12.4	10.8	11.6	10.2	Ø.	<b>%</b>	®•¢	11.3	11.3	13.8	11.4
TUR- 810- 1TY (NTU)	;	1	1	;	1	1	1	1	1	1	;	:	;	1	1	1	;	!	!	!	2.0	1.0
TUR- BID- ITY (JTU)	7		1	ហ	٢	'n	N	2	~	4	6	~	<b>~</b>	-	E	3	9	o	en	M	:	!
COLOR (PLAT- INUM COBALT UNITS)	;	:	1	18	38	10	04	ເກ	12	15	15	n	~	10	20	30	νo	10	35	55	02	30
TEMPER- ATUKE, WATER (DEG C)	7.5	13.5	21.0	20.0	20.5	19.0	7.0	3.0	1.5	0.	1.0	0.4	15.0	11.0	20.0	22.0	19.5	17.0	11.0	13.0	0.4	10.0
TEMPER- ATURE, AIR (DEG C)	2°5	22.0	25.0	19.0	30.0	18.5	3.5	4.0	0.	6.5	0.6	21.0	25.5	16.5	52.5	22.0	23.0	12.5	14.0	14.0	0.0	<b>5</b> •¢
PH (UNITS)	8.2	2.5	7.8	8.3	7.5	7.8	7.7	7.9	8.0	7.6	d.5	7.7	8.6	8.0	7.7	7.8	7.8	7.5	7.5	7.3	7.9	8.0
SPE- CIFIC CON- DUCT- ANCE (MICRO- MHOS)	251	210	251	064	165	371	219	545	370	420	450	554	280	232	380	381	375	552	529	180	172	195
STREAM- FLOW. INSTAN- TANEOUS (CFS)	27	30	15	3.2	22H	4.9	36	13	15	10	1.1	101	30	35	5.9	1.6	2.7	7.6	28	100	130	23
T I ME	1976 0935	9860	1005	1125	1340	1340	1130	1040	1450	1025	1030	1030	1035	1030	1115	1200	1030	1030	1035	1100	1100	1100
DATE	APR , 19	10	22.	13	11	21	27.	22.	:	26	23.		19.	111.	22	200 200 200 200 200 200 200 200 200 200	23.	21	19.	:	• :	APR , 19

Table 11. -- Water-quality analyses. -- continued

Site 2, Norwalk River at Georgetown, Conn.--Continued

SOLIDS. SUM OF CONSTI- TUENTS, DIS- SOLVED	;	;	;	529	45	200	119	151	182	502	23A	105	140	125	210	751	757	136	137	108	105	130
SOLIDS, RESIDUE AT 180 DEG, C DIS- SOLVED (MG/L)	143	132	167	335	102	240	135	161	<b>412</b>	232	683	133	171	11	780	245	212	159	159	101	108	166
SILICA. DIS- SOLVED (MG/L AS SIO2)	1	:	;	3,3	6.9	4.2	10	7.7	8. 8.	10	6.3	6.5	3.3	6.1	5.5	5.4	7.5	5.2	10	8.0	7.1	3.5
CHLO- RIDE, UIS- SOLVED (MG/L AS CL)	25	16	23	30	13	14	2	52	19	99	29	92	32	25	55	1 1	7	22	25	18	54	25
SULFATE DIS- SOLVED (MG/L AS SO4)	;	!	i	643	14	35	17	25	97	62	36	Ŧ	21	16	#	74	74,	<b>ئ</b> ر	62	11	11	90
CARBON DIOXIDE DIS- SOLVED (MG/L AS CO2)	۲.	1.4	2.2	σ.	3.0	2.4	2.4	1.9	1.3	3.7	5•	1.1	€.	1.2	5.6	2.8	3.0	3.7	3.2	5.1	σ.	1.2
ALKA- LINITY (MG/L AS CACO3)	56	56	7.1	18	7,7	61	61	16	65	75	83	30	19	62	19	0	ž	09	53	53	39	62
CAR- BONATE (MG/L AS CO3)	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	э	э	9	0	0	0
BICAR- BONATE (MG/L AS HC03)	89	99	86	106	9	96	74	66	2	76	101	36	74	75	82	110	120	7.3	49	79	14	15
SODIUM AD- SORP- TION RATIO	1976	• 5	\$	•	4	r.	4	• 5	5.		æ	•	2.	9.	9.	9.		• 5	r.		9.	
DATE	APR . 19	MAY 10.	21.	13	11.	21	27	22.5	29.	26	23	A 0.0	10.	11	22	25	23	21	19.	10	-	APK , 19

Table 11. --Water-quality analyses. -- continued

Site 2, Norwalk River at Georgetown, Conn. -- Continued

NITRO- GEN+ TOTAL (MG/L AS NO3)	2.7	2.3	4.6	• •	3.7	4.5	2.2	5.9	4.6	4.3	6.6	3.5	2.7	1.4	3.7	<b>†:</b> 1	α·1	3.2	3.4	3.5	0.4	2.7
NITRO- GEN+ TOTAL (MU/L AS N)	. 40	.53	1.0	1.0	83	1.0	64.	• 65	1.0	7.1	7.1	.78	.61	04.	.84	26.	1.8	.72	.76	. 7B	2.	.60
NITRO- GEN, AM- MONIA + ORGANIC TOTAL (MG/L AS N)	.38	.28	.50	09.	.60	.70	•33	\$5.	4.	1:1	1.1	• 30	04.	*25	<b>†</b>	90.	1.3	.45	<b>†</b>	• 55	. 2A	.41
NITRO- GEN, ORGANIC TOTAL (MG/L AS N)	35	!	64.	.51	.53	. 45	۶۶.	• 40	35	.83	69.	<b>47.</b>	.31	.17	46.	.55	1.1	.34	04.	.54	٤٤.	.41
NITRO- GEN, AMMONIA UIS- SOLVED (MG/L AS NH4)	<b>*</b> 0 •	}	.01	.12	60.	.32	• 00	• 0 •	.17	.35	.53	90.	.12	•00	.13	<b>*</b> 0 •	•26	.14	.05	.01	• 00	00.
NITHO- GEN, AMMONIA DIS- SOL VED (MG/L AS N)	• 03	}	• 01	60.	.07	•25	• 05	• 05	•13	.27	.41	• 0•	• 00	• 05	• 10	•03	• 20	.11	*0	.01	• 05	00.
NITHO- GEN+ NOZ+NO3 TOTAL (MG/L AS N)	.22	•25	•53	.43	.23	•31	•16	0 4 •	•56	1.0	66*	84.	.21	.18	4.	• 34	.53	.27	•32	.23	.63	.14
SOLIDS. RESIDUE AT 105 DEG. C. TOTAL (MG/L)	146	148	174	382	117	556	138	167	234	248	594	146	173	29	272	315	280	187	188	127	130	179
SOLIDS. DIS- SOLVED (TONS PER DAY)	10.4	10.7	6.85	2.89	62.8	4.15	13.1	5.65	8.59	6.58	8.41	36.3	13.9	7.28	2.19	1.23	1.98	3.26	12.0	28.9	37.9	10.3
SOLIDS. DIS- SOLVED (TONS PER AC-FT)	976	.18	.23	44.	.14	•33	.18	.22	65.	.32	• 38	.18	.23	.10	•38	•39	.37	.22	• 22	.15	.15	•23
DATE	APR . 1	10.	201 21.:	13	11.	21	22	22.0	•	. :	23.	500		:	22.	25. 25.	829 839	21	19.	-	22	19.

Table 11.--Water-quality analyses.--continued Site 2, Norwalk River at Georgetown, Conn.--Continued

SEDI- MENT DIS- CHARGE, SUS- PENDED	62*	•5•	.21	90.	3.7	.07	• 39	.11	.16	.14	.12	4.6	1.9	32	• 0 •	• 0 9	•20	.23	• 60	.81	1.3	.19
SEDI- MENT, SUS- PENDED (MG/L)	4	e	ហ	1	•	4	4	e	4	ß	4	1.7	23	336	'n	18	28	1.1	x	e	īv	3
METHY- LENE BLUE BCTIVE SUB- STANCE (MG/L)	00.	00.	00•	.10	00.	• 00	00.	00.	00.	.10	.10	00.	00.	:	:	ł	;	;	05.	!	00.	00.
CARBON. ORGANIC TOTAL (MG/L AS C)	3.1	13	۲.	9.0	٠ ٠	6.2	m • •	\$ &	<b>L</b> .4	3.6	3.3	3.7	3.4	٥.5 ک	4.2	5.3	7.6	6.9	ð.6	10	0 • 4	8.1
ZINC. DIS- SOLVED (UG/L AS ZN)	20	0 7	0,	40	20	9	30	1	9	160	30	30	30	30	20	30	0	0 7	.04	30	0 1	0 7
MANGA- NESE+ DIS- SOLVED (UG/L AS MN)	20	;	20	0 7	30	30	30	40	9	9	90	30	0 7	40	30	20	06	70	0 7	70	07	0 7
LEAD. UIS- SOLVED (UG/L AS PB)	14	;	σ	23	21	55	60	17	13	7	~	v	25	77	10	~	15	77	15	11	<b>3</b> 0	S
IKON. DIS- SOLVED (UG/L AS FE)	150	200	320	9	180	200	300	160	160	80	0 7	120	360	390	170	120	100	180	9	230	220	180
COPPER, DIS- SOLVED (UG/L AS CU)	9	0	э	10	9	0	0	0	0	0	0	9	0	10	٥	11	•	4	ฑ	٣	4	4
PHOS- PHORUS. TOTAL (MG/L AS P)	1976	060.	.230	.120	.180	.100	.110	.060	.050	1977	.110	.040	.040	.100	.100	.140	.220	.190	.110	.120	1978	010.
DATE	APR . 19	MAY 10	JUN 21	JUL 13	AUG 11	SEP 21	0CT	NOV 22	DEC 29	JAN . 1	FF8 23	MAR 29	APR 19	MAY 11	JUN 22	JUL 25	AUG 23	21	0CT 19	:	MAR . 1	19

Table 11. -- Water-quality analyses. -- continued Site 3, Norwalk River at Cannondale, Conn.

Table 11.--Water-quality analyses.--continued Site 3, Norwalk River at Cannondale, Conn.--Continued

SODIUM. DIS- SOLVED (MG/L AS NA)	11	0.1	4.0	20	7.3	13	7.6	12	12	19	23	12	14	12	17	17	20	12	12	x.0	12	13
MAGNE- SIUM, DIS- SOLVED (MG/L AS MG)	6.8	υ •	7.3	4.4	8.4	8	6.8	30 0	8.0	ø	80	5.4	7.6	1.2	8.5	01	==	9.9	1.6	3.6	5.3	5.6
CALCIUM DIS- SOLVED (MG/L AS CA)	23	11	17	ıç	11	36	20	62	33	34	£ <b>7</b>	61	52	54	43	4 ]	39	30	ς <u>ς</u>	18	17	19
HARD- NESS* NONCAR- BUNATE (MG/L CACU3)	31	13	32	18	7	4	5	36	51	٥٠	67	\$	24	33	9	59	9	55	<b>4</b> 3	42	2	34
HARD- NESS (MG/L AS CACO3)	85	99	16	170	62	130	78	110	120	120	140	7.0	*	06	140	140	140	100	46	6.8	40	7.1
STREP- TOCOCCI FECAL: (COLS: PER 100 ML)	21	69	19	220	2900	800	;	;	;	:	;	;	;	;	;	:	:	;		;	1	!
STREP- TOCOCCI FECAL, KF AGAR (COLS, PER 100 ML)	;	;	;	;	;	:	130	31	13	20	81500	160	0 7	34	200	1200	160	1200	K5000	00099	32	25
COLI- FORM, FECAL, 0.7 UM-MF (COLS./	:	:	;	;	;	;	220	130	63	260	19	30 30	270	330	1300	2500	800	2400	006	3600	ю 10	220
COLI- FORM: FECAL: 0.45 UM-MF (COLS:/	06	610	1100	1200	1900	1400	;	;	;	:	;	;	;	1	;	:	;	;	:	;	:	;
COLI- FORM, TOTAL, IMMED. (COLS, PER	1976 180	1100	3400	0007	0006	00077	780	330	310	1311	320	200	;	24000	3000	10000	2000	4000	4800	5	976 HB	480
DATE	- :	10.	20N	13	111	21	27.	25. 22.	29	• •	23	200 200	19.	111	22.	25	23	21	19.	-	┛.	APP .

Table 11.--Water-quality analyses.--continued

Contir	
Conn	
Cannondale	
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Site	

SOLIDS. SUM OF CONSTI- TUENTS. DIS- SOLVED	ì	1	1	522	66	177	119	150	165	193	228	113	134	123	201	198	210	141	137	104	103	110
SOLIDS. RESIDUE AT 180 UEG. C UIS. SOLVED (MG/L)	145	125	165	997	101	502	132	160	182	212	792	130	159	154	555	261	253	186	166	104	110	145
SILICA. DIS- SOLVED (MG/L AS SIO2)	1	;	i	1.3	7.1	4.5	10	7.4	8.60	10	6.8	9.9	3.4	2.0	5.5	1.4	x •	5.3	10	8.0	7.2	4.5
CHLO- RIDE, UIS- SOLVEU (MG/L AS CL)	23	19	23	55	13	37	. 19	27	36	54	19	27	32	23	25	74	<b>5</b>	32	23	19	*7	52
SULFATE DIS- SOLVED (MG/L AS SO4)	;	1	1	4.1	77	30	17	22	52	27	37	51	21	16	37	34	<b>1</b>	32	62	18	16	20
CARBON DIOXIDE DIS- SOL VEU (MG/L AS CO2)	π.	1.6	2.0	<b>?•</b>	3.0	3.1	5.9	•	2.0	3.3	20	2.0	7.	.,	1.9	1.9	1.3	2.4	5.5	5.1	1.4	٠ ب
ALKA- LINITY (MG/L AS CACO3)	ų 4	53	99	79	84	61	65	7.2	40	x o	11	0,4	55	57	62	47	82	<b>x</b>	14	4.2	35	37
CAR- BONATE (MG/L AS CO3)	0	0	0	0	0	0	0	0	0	0	9	0	0	0	0	0	0	0	э	Э	0	0
BICAR- BONATE (MG/L AS HC03)	99	9	90	96	59	96	12	88	78	83	76	7 1	63	69	7.5	96	100	65	29	51	43	4,
SODIUM AD- SORP- TION HATIO	1976	5.	~5		4.	ស្	<b>5</b> •	٠,	3.	8. 1977	α.	•	•	•	•	¢.		\$.	5.	5.	1978	
DATE	APR . 1	10	21	13	11.	21	27	25. 22.	29	JAN . 1	23.	29	194	11.	JUN 22.	35. 25.	23	21		:	- :	14.

Table 11. -- Water-quality analyses. -- continued

Site 3, Norwalk River at Cannondale, Conn. -- Continued

NITHO- GEN. TOTAL (MG/L AS NO3)	2.1	5.4	5.4	6.5	3.5	3.4	2.3	5.6	3.8	8.7	13	3.0	3.3	2.3	3.4	3.0	4.9	5.5	3.2	4.2	• • 0	3.3
NITHO- GEN+ TOTAL (MG/L AS N)	.47	.54	1.0	99.	. 40	.76	.52	Ac.	.85	٥٠٧	۷.۶	.08	.75	.51	30	.67	1.5	1.3	.72	56.	.91	74.
VITRO- GEN•AM- MONIA • OMGANIC TOTAL (MG/L AS N)	•23	•33	• •	.45	. 55	.45	• 33	.18	.28	16.	1.9	٠٥٥	.50	.31	07.	.52	70.	.91	.39	.72	.31	•36
NITRU- GEN• ORGANIC TOTAL (MG/L AS N)	.14	.30	94.	77.	4.	54.	٤٥.	1	.21	٠٢.	1.6	.15	.45	<b>.</b> 8	.37	64.	8	٠٢.	.36	.71	.27	• 35
NITRO- GEN• GEN• AMMONIA DIS- SOLVEU (MG/L	\$0.	•0•	.03	.01	60.	• 04	\$0.	• 05	60.	.28	.35	90•	•0•	<b>*0</b> •	<b>*</b> 0 •	*0*	*0*	.15	• 0 •	.01	<b>.</b> 05	• 01
NITRO- GEN• AMMONIA DIS- SOLVED (MG/L AS N)	• 0 •	.03	70.	.01	.07	.03	*0*	• 0 •	.07	.22	.27	• 05	• 05	.03	.03	.03	.03	.12	.03	.01	<b>†</b> 0.	• 0 1
NITRO- GEN+ NOZ+NO3 TUTAL (MG/L AS N)	*5*	.21	.54	.21	• 25	18.	•19	04.	.57	66.	1.0	4. 84	•25	.20	8	.15	.61	.34	.33	.23	.60	36.
SOLIDS. RESIDUE AT 105 DEG. C. TOTAL (MG/L)	150	128	172	321	121	218	141	168	200	234	462	135	163	152	524	274	285	216	186	126	125	152
SOLIDS. DIS- SOLVED (TONS PER DAY)	11.0	10.5	7.13	2.44	65.2	3.71	13.2	5.75	7.86	6.30	8.55	37.2	13.7	15.4	5.09	1.20	1.91	4.01	13.2	29.5	40.5	6.45
SOLIDS. DIS- SOLVED (TONS PER AC-FT)	1976	.17	.22	.36	.14	.28	.18	.22		62.	.36	.18	•25	.21	.35	.36	.34	•25	.23	4I.	.15	07.
OATE	APR . 19	10.	20N	13	11	21		25.	29	26	23.	29.	5.0	111.	22	25.	23.	21	161	•	. :	19.

Table 11.--Water-quality analyses.--continued

Site 3, Norwalk River at Cannondale, Conn.--Continued

DATE	PHOS- PHORUS: TOTAL (MG/L AS P)	COPPER, DIS- SOLVED (UG/L AS CU)	IRON; DIS- SOLVED (UG/L AS FE)	LEAD, DIS- SOLVED (UG/L AS PB)	MANGA- NESE, DIS- SOLVED (UG/L AS MN)	ZINC, DIS- SOLVED (UG/L AS ZN)	CARBON, ORGANIC TOTAL (MG/L AS C)	METHY- LENE BLUE ACTIVE SUB- STANCE (MG/L)	SEDI- MENT, SUS- PENDED (MG/L)	SEDI- MENT DIS- CHARGE: SUS- PENDED (T/DAY)
APR • 1										
12 May	•050	0	120	4	40	20	3.7	.00	3	•23
10	.080	0	150	7	20	10	4.5	.00	3	.25
JUN 21	•200	0	290	16	40	50	9.2	•00	4	.17
JUL										
13 AUG	•080	10	40	9	10	10	6.3	.10	8	• 07
11	.180	0	190	9	20	20	9.1	•00	11	7.1
SEP 21	.080	0	210	16	10	50	6.9	•00	1	•02
OCT	000	0	200	25	20	30	7.0	0.0	_	20
27 NOV	.080	0	300	23	30	30	7.0	•00	2	.20
22 DEC	•060	0	140	37	30	40	5.0	•00	3	.11
29	•050	0	130	27	40	50	3.7	•00	3	•13
JAN , 1 26	977 •100	0	50	8	50	140	4.0	•00	4	.12
FEB										
23 MAR	•220	0	40	5	40	40	4.0	.10	5	.16
29	.080	0	120	5	20	40	5.7	.00	46	13
APR 19	•040	0	320	18	30	30	4.0	•00	21	1.8
MAY		_								
11 JUN	.100	3	370	35	40	30	7.1		52	5•2
22	.080	5	100	22	0	30	4.6		12	.10
25	•090	10	110	10	20	10	6.0		162	.74
AUG 23	•310	6	100	15	20	20	7.8		7	•05
SEP	•310		100	13	20	20	7.0	-	,	•03
21 OCT	.160	5	140	23	30	40	7.6		18	•39
19	.100	3	10	26	30	40	7.9	•00	7	•56
NOV 10	.120	4	220	59	30	30	9.3		2	•57
MAR , 1	978									
22 APR ,	•040	2	100	5`	30	30	3.3	•00	2	•74
19	.030	3	360	7	30	50	7.8	.00	81	5.3

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## Table 11.--Water-quality analyses.--continued Site 4, Comstock Brook at North Wilton, Conn.

			Dice	., сомыс	OCK DIOOK	at Notth	WIICOII,	com.			
DATE	TIME	STREAM- FLOW, INSTAN- TANEOUS (CFS)	SPE- CIFIC CON- DUCT- ANCE (MICRO- MHOS)	PH (UNITS)	TEMPER- ATURE, AIR (DEG C)	TEMPER- ATURE, WATER (DEG C)	TÜR- BID- ITY (JTU)	OXYGEN, DIS- SOLVED (MG/L)	OXYGEN, DIS- SOLVED (PER- CENT SATUR- ATION)	COLI- FORM. TOTAL. IMMED. (COLS. PER 100 ML)	COLI- FORM, FECAL, 0.45 UM-MF (COLS./
APR	1205	0.0	126	7.5	17.0	10.5	1	11.4	- 102	20	2
13 MAY	1325	8.0	124								1
11 JUN	1400	6.3	108	7.5	19.5	15.0	1	10.5	103	230	31
23	1240 1025	1.9	124	7.4	29.5	21.0	1 	8.4	93	330	63
DATE	STREP- TOCOCCI FECAL, (COLS. PER 100 ML)	HARD- NESS (MG/L AS CACO3)	HARD- NESS: NONCAR- BONATE (MG/L CACO3)	CALCIUM DIS- SOLVED (MG/L AS CA)	MAGNE- SIUM, DIS- SOLVED (MG/L AS MG)	SODIUM, DIS- SOLVED (MG/L AS NA)	SOUIUM AD- SORP- TION RATIO	BICAR- BONATE (MG/L AS HCO3)	CAR- BONATE (MG/L AS CO3)	ALKA- LINITY (MG/L AS CAC03)	CARBON DIOXIDE DIS- SOLVED (MG/L AS CO2)
APR		_						2.1	•	23	1
13 May	3	35	12	8.7	3.2	6.2	.5	28	0		1.4
11 JUN	53	31	6	7.5	2.9	5.6	.4	30	0	25	1.5
22	74	44 	7	11	3.9	5.5 	.4	44	0	36 	2.8
DATE	CHLO- RIDE, DIS- SOLVED (MG/L AS CL)	SOLIDS, RESIDUE AT 180 DEG. C DIS- SOLVED (MG/L)	SOLIDS, DIS- SOLVED (TONS PER AC-FT)	SOLIDS, DIS- SOLVED (TONS PER DAY)	SOLIDS, RESIDUE AT 105 DEG. C, TOTAL (MG/L)	NITRO- GEN+ NO2+NO3 TOTAL (MG/L AS N)	NITRO- GEN. AMMONIA DIS- SOLVED (MG/L AS N)	NITRO- GEN• AMMONIA DIS- SULVED (MG/L AS NH4)	NITRO- GEN; ORGANIC TOTAL (MG/L AS N)	NITRO- GEN+AM- MONIA + ORGANIC TOTAL (MG/L AS N)	NITRO- GEN+ TOTAL (MG/L AS N)
APR		4.0	0.0	1 24	91		0.1	.01	.19	.20	.32
13 MAY	12	62	.08	1.34	81	.12	.01			.35	.44
11 JUN	8.9	69	•09	1.17	69	•09	.02	.03	.33		•69
23	9.7	88	•12	.45 	88 	.29		.03	.38	.40	
DATE	NITRO- GEN, TOTAL (MG/L AS NO3)	PHOS- PHORUS, TOTAL (MG/L AS P)	COPPER. DIS- SOLVED (UG/L AS CU)	IRON, DIS- SOLVED (UG/L AS FE)	LEAD. DIS- SOLVED (UG/L AS PB)	MANGA- NESE • DIS- SOL VED (UG/L AS MN)	ZINC, DIS- SOLVED (UG/L AS ZN)	CARBON+ ORGANIC TOTAL (MG/L AS C)	METHY- LENE BLUE ACTIVE SUB- STANCE (MG/L)	SEDI - MENT + SUS - PENDED (MG/L)	SEDI- MENT DIS- CHARGE, SUS- PENDED (T/DAY)
APR 13	1.4	.020	0	70	7	20	0	3.9	•00	5	.11
MAY				100	9	10	0	1.9	•00	2	.03
JUN	1.9	.030	0				0	11	•00	43	.22
23	3.1	• 050 ••	0	430	10	40		***			

Table 11.--Water-quality analyses.--continued Site 5, Bryant Brook at Wilton, Conn.

COL 1- FORM. TOTAL. IMMED. (COLS. PEH	S.	4300	2100	CAK- HONATE (MG/L ASVEC3)	•	9	۰:	14.544.5	82	.23	33		00	00.	01
OXYGEN. DIS- SOLVED (PER- CENT SATUK- ATION)	104	104	101	dicak- bonate (MG/L AS.ch	. 91	ž	å !	NITPO- CEN.AM- MONIA . OVGA-IC TOTAL (MC/L AS '-)	•	•	•	MFT4Y- LENE 9LUE ACTIVE SUH- STANCE (MG/L)	•	•	•
	2.5	5.0	0:0	· ·	٠	ę.	۲: ا	NITHO- GEN+ NOZ+NO3 TOTAL (MG/L	¥	.55	.37	CAPHON, ORGANIC TOTAL (MU/L AS C)	1.6	2.6	4:
OXYGEN. DIST SOLVED (MG/L)	==	~		SODIUM ADDI SOUPE TION				SOLIUS, RESIDUE AT 105 P DEG. C. TOTAL (46/L)	81	10	٤:	21NG. D1S- SOLVED (UGAL	9	•	2 !
10k 111 111	~	-	<b>≈</b> ;	SOUTUM. UIS. SOLVEU (MG/L AS MA)	a a	8.3	7.7			.37	<b>9</b> !		<b>5</b> 0	2.	21
TEMPEH- ATURE, WATER (DEG C)	8.5	16.0	25.0	MAGRE- SIUM. S DIS- SOLVED (MG/L AS MG)	2.3	9.7	9:0	SOLIDS. DIS- SOLVED (TONS PEA DAY)	•	·;	5:	MANCA- NFSE, DIS- SOLVED (UG/L AS MN)		_	
	7.0	ر. د	9 !		•	۵.4	:	SOLIDS. DIS- SOLVED (TONS PEX AC-FT)	=		.12	IRON. DIS- SOLVED (UG/L AS FE)	9	80	150
TEMPER- ATUME, AIR (DEG C)	7,	*,	29.0	CALCIUM DIS- SOLVED (MG/L AS CA)	*	œ.	Ξ.		ã	ą,	ş ¦	UDPER. 1 UIS- SOLVED 5 (UG/L AS CU)	0	0	•;
rd (ONITS)	7.1	7.5	7.6	HARU- NESS. NONCAM- HONATE (MG/L CACU3)	67	10	٠ ;	SOLIDS. HESIDUE AT 180 DEG. C DES. SOLVED (MG/L)				3	-	_	01
•	132	123	<u> </u>	HAPD- PRINCESS NICK NG/L BC	8	33	9	CHLO- RIUE. 015- SOLVEU (MG/L AS CL)	1	12	21	PHOS- PHORUS: TOTAL (MG/L AS P)	. 020	.030	
			•					CAMBON DIOXIDE 01S- SOLVED (MG/L AS CO2)	2.0	·:	1.5	N17RO- GEN+ TOTAL (MG/L	7.2	2.1	3.1
STREAM- FLOW, INSTAN- TANEOUS (CFS)	1.6	1.6	ę. ;	STREP- TOCOCCI FECAL+ (COLS+ PER- 100 ML)		5.	B220		13	23	ī ;	4	.62	97.	
7 I ME	0641	1350	1410	COLI- FORM. FECAL. 0.45 UM-MF CONS./	~	10	360	ALKA- LINITY (M5/L AS CACO3)		-		NITRO- GEN. 101AL (MG/L AS N)		•	•
	:	:			:	:	::	UATE	APR 12	10	JUN 22.:.	UATE	12	10.	22. 23.
DATE	APR 12.	2	22.2	UATE	APR 12	0.7	22		•	•	•		•	•	•

Table 11. -- Water-quality analyses. -- continued

Conn.	
ŏ	
Wilton,	
South	
at	
River	
Norwalk	
9	
Site	

OXYGEN. DIS- SOLVED (PER- CENT SATUR-	1117	119	104	112	16	100	101	117	109	104	105	101	105	100	115	114	106	113	104	102	103	103
OXYGEN. DIS- SOLVED (MG/L)	14.3	12.4	9.5	10.4	0.6	9.5	12.4	15.5	15.6	14.4	14.2	11.2	10.4	10.6	10.4	10.4	9.6	11.2	11.6	10.8	13.2	12.0
105 110- 117<	;	:	ł	i	:	;	;	:	:	;	;	;	1	:	;	;	:	:	:	:	2.0	<b>.</b>
108- 117- 110-	1	-	~	~	9	8	2	-	-	8	٣	Ю	~	7	2	-	'n	4	~	4	;	;
COLOM (PLAT- INUM COBALT	;	:	:	e	35	1	38	10	10	~	~	~	ហ	0.1	20	0.1	20	151	17	37	2	20
TEMPER- ATURE, WATER (DEG C)	7.0	14.0	22.0	19.0	19.0	18.0	6.5	0.4	1.0	2.0	3.0	11.0	16.0	13.0	21.0	20.5	20.0	16.0	11.0	13.0	5.0	9.0
TEMPER- ATURE, AIR (DEG C)	7.0	23.5	28.0	22.5	26.5	55.5	0.9	5.0	\$	1.5	15.0	28.5	54.5	17.5	26.0	22.5	25.0	15.5	14.0	16.5	11.0	<b>9.</b> 0
PH (UNITS)	15.	8.2	1.6	7.8	7.5	7.4	7.5	3.8	1.5	7.5	7.8	7.5	8.5	8.2	8.0	7.2	7.8	7.1	7.3	7.2	7.5	7.4
SPE- CIFIC CON- DUCT- ANCE (MICRO- MHOS)	220	189	236	265	146	235	198	240	27.1	380	330	210	217	555	550	275	568	215	220	170	- 134	165
STREAM- FLOW• INSTAN- TANEOUS (CFS)	74	ě	21	7.8	262	71	16	23	52	15	90	165	23	90	7.6	3.8	5.5	23	45	210	572	99
I E	1976 1210	1235	1310	1455	1100	1120	1345	1350	1130	1300	1300	1330	1330	1400	1330	1400	1300	1310	1400	1400	1400	1300
DATE	• :	10	23	13	111.		22.	22.	:	26 26	23.	25. 65.	19.	11	22.	255. A16	23	21	19	:	· : ,	19.

Table 11. -- Water-quality analyses. -- continued

		571	rable 11.	Table 11 Water-quality analyses continued	ality and	lyses	continued			
		Site 6	6, Norwalk	River at	South	Wilton, Cor	Conn Continued	nued	٠	
DATE	COLI- FORM: TOTAL: IMMED: (COLS: PER:	COLI- FORM. FECAL. 0.45 UM-MF. (COLS./	COLI- FORM. FECAL. 0.7 UM-MF (COLS./	STREP- TOCOCCI FECAL. KF AGAR (COLS. PEH	STREP- TOCOCCI FECAL, (COLS, PER 100 ML)	HARD- NESS (MG/L AS CACO3)	HARD- NESS. NUNCAR- BONATE (MG/L CACO3)	CALCIUM UIS- SOLVEU (MG/L AS CA)	MAGNE- SIUM. DIS- SDLVED (MG/L AS MG)	SODIUM. DIS- SOLVED (MG/L AS NA)
APR . 1	1976 470	22	;	;	<b>5</b>	61	<u> </u>	16	5.1	-
MA∀ 10	2200	7.0	;	;	82	57	18	15	3 a	10
23.	2300	160	1	:	88	85	35	54	6.2	12
JUC 13	00099	220	;	;	16	93	υ Θ	56	6.9	20
11:	10000	880	1	;	3300	53	7	14	4.3	7.1
21.5	1900	450	;	!	120	86	30	23	7.0	12
22.	850	;	190	88	;	99	<b>5</b> 0	1.7	5.6	0.6
22.	1100	;	27	4	:	86	34	23	7.0	11
:	34000	;	160	53	:	91	38	92	6.3	12
: :	240	;	30	30	;	66	1,	2 k	5.1	52
	28000	:	40	52	;	45	<b>‡</b>	92	7.2	21
62	260	;	20	9	;	20	11	1.1	1.8	13
	130	;	84	12	:	9	92	1.7	5.5	12
111.	18000	;	58	16	:	83	18	23	6.3	12
	36000	;	730	34	1	9	3 3	52	6.1	15
25. 25.	3000	;	530	700	:	86	55	87	6.8	16
200	9600	;	540	450	:	86	47	7.2	7.5	16
21	4400	;	3100	1900	;	78	<b>3</b> 7	22	5.7	12
19.	4400	;	230	200	;	87	47	#2	6.5	12
:	6000	:	1700	2000	;	57	21	ĬŽ	4.8	9.6
. :	20	i	14	21	:	64	17	13	4.1	
19	1200	1	100	#	!	23	10	5.8	2.1	7.4

Table 11.--Water-quality analyses.--continued

Site 6, Norwalk River at South Wilton, Conn. --Continued

SOLIDS. SUM OF CONSTI- TUENTS. DIS- SOLVED	;	:	;	144	83	132	107	123	141	181	161	102	107	117	138	150	151	124	131	96	06	L4
SOLIDS. RESIDUE AT 180 DEG. C DIS- SOLVED (MG/L)	119	119	120	166	45	138	121	171	159	208	184	117	124	133	178	202	188	151	156	96	SA	88
SILICA. DIS- SOLVED (MG/L AS SI02)	;	;		0.4	7.1	6.3	10	9.9	0.5	0.5	7.9	7.1	1.	6.3	6.9	4	7.1	7.6	3.0	4.8	7.1	7.0
CHLO- RIDE. DIS- SOLVED (MG/L AS CL)	22	51	62	31	12	58	50	52	33	58	9	92	27	5	33	24	35	52	56	18	20	ю 60
SULFATE DIS- SOLVEU (MG/L AS SO4)	1	;	i	25	15	21	11	61	23	54	22	17	11	1.1	54	. 22	27	33	28	81	15	7.6
CARBON DIOXIDE DIS- SOLVED (MG/L	۶.	3.	2.5	1.7	5.4	1 .	2.8	4	3.2	3.2	1.6	2.0	• 5	9.	<b>5.</b>	5.2	1.6	4.7	3.9	4.	2.0	1.0
ALKA- LINITY (MG/L AS CACO3)	38	39	90	99	39	57	9	53	53	53	51	33	36	4	46	43	52	30	0 1	36	33	13
CAP- BONATE (MG/L AS CO3)	э	0	0	9	0	0	Э	Э	9	0	0	9	0	0	0	9	9	0	0	0	0	0
BICAK-BONATE (MG/L AS	9	4 30	61	<b>9</b>	47	69	56	49	49	49	29	0 7	4 20	99	99	25	63	37	54	4	9	16
SODIUM A0- SORP- TION PATIO	1976	•	9.	•	4	9.	5.	\$.	.5	1:1	5.	æ	¢.	•			.,	ę.	9.	9.		٠.
DATE	APR . 10	10	23.	13	11	21	200	22.	•	•	23.	200		11.	22	25°	23.	21	19	:	•	APR , 19

Table 11. -- Water-quality analyses. -- continued

Site 6, Norwalk River at South Wilton, Conn. -- Continued

NITRO- GEN• TOTAL (MG/L AS NO3)	2.3	2.3	4.5	5.6	3.5	3.1	2.2	2.1	3.7	1.9	6.1	5.9	3.9	5.9	7.	3.9	4.3	5.4	3.4	6.9	3.6	1.4
NITRO- GEN+ TOTAL (MG/L AS N)	.51	.53	1.0	.58	.80	. 71	05.	14.	* * *	1.8	1.4	• 65	.87	• 65	66.	.87	÷.	1.2	.76	1.0	. 42	.31
NITRO- GEN•AM- MONIA • OKGANIC TOTAL (MG/L AS N)	. <b>2</b> 0	• 25	.35	.33	•55	•30	• 30	•15	•20	.85	.54	•10	.60	.37	24.	04.	• 55	.78	35.	1.3	.28	.13
NITRO- GEN+ OKGANIC TOTAL (MG/L AS N)	;	:	:	.28	• 50	•26	.27	•13	81.	.76	54.	.07	• 58	.36	38	.30	.53	99	.33	1.3	12.	.13
NITRO- GEN+ AMMONIA DIS- SOLVEU (MG/L	;	:	:	90.	•0•	• 05	• 0 •	.03	£0.	.12	• 00	*0.	•03	• 01	\$0.	.13	.03	.13	• 03	00•	.01	00.
NITRO- GEN+ AMMONIA DIS- SOLVED (MG/L AS N)	;	;	;	<b>50</b> •	• 05	• 0 •	• 03	• 02	• 02	60*	• 05	• 03	• 02	.01	*0.	.10	• 05	•10	• 05	00•	.01	00.
NITRO- GEN+ NOZ+NO3 TOTAL (MG/L AS N)	.31	•28	19.	ځځ.	•25	.41	• 20	• 32	.64	<b>76.</b>	<b>78.</b>	• 55	.27	.28	.57	14.	.41	54.	.41	•26	<b>\$</b> C.	.18
SOLIDS. RESIDUE AT 105 DEG. C. TOTAL (MG/L)	128	124	168	180	122	154	127	183	173	225	199	125	127	144	174	237	197	167	111	125	101	52
SOLIDS. DIS- SOLVED (TONS PER DAY)	23.8	21.8	6.80	3.50	65.1	5.22	24.8	8.14	10.7	8.42	9.94	52.1	19.1	28.7	4.52	2.12	2.79	9.75	19.0	54.4	51.6	10.5
SOLIDS. DIS- SOLVED (TONS PER AC-FT)	1976	.16	.16	.23	.13	•19	.16	.18	.22	.28	.25	.16	.17	.18	•2•	.28	•26	.21	.21	.13	.12	90.
DATE	APR . 19	10.	23.::	13	11	21	27.	22.00	29	255	23.	29.	19.	111	22	25.	23	21	19.	10		14

Table 11. -- Water-quality analyses. -- continued

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	. Conn Continued
	_
,	ilton,
)	at South W
-	at
	River
	Norwalk
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	_

SEUI- MENT DIS- CHARGE. SUS- PENDED	07.	, 55	.45	.04	11	• 0 •	.41	•19	•13	90•	• 16	6.4	χ •	7.1	•33	• 64	.13	.87	16.	0.1	<b>5</b> .5	69	41
SEGI- MENT. SUS- PENDED	~	<b>6</b>	æ	4	15	-	<b>?</b>	m	2	~'	E	11	55	33	13	62	•	14	8	^	4	389	4.
METHY- LENE BLUE ACTIVE SUB- STANCE (MG/L)	00•	00.	00.	00•	• 00	00.	• 00	00.	00.	.10	.10	00.	00.	;	;	;	ł	ŀ	00.	;	00.	00.	
CARBON. ORGANIC TOTAL (MG/L AS C)	2.4	3.1	6.1	4.0	0.6	3.1	1.4	1.4	1.4	3.6	3.3	6.1	3.0	6.2	5.1	5.6	1.9	9.1	6.1	11	<b>8.</b> 0	5.2	- j
ZINC. DIS- SOLVED (UG/L AS ZN)	10	50	20	10	10	10	50	0	0,4	9	30	92	10	50	10	10	0.Z	20	30	20	10	20	Ī
MANGA- NESE. I DIS- SOLVED (UG/L AS MN)	30	20	30	10	50	20	, 20	50	0 1	40	20	٥٧	20	50	0	30	30	0 7	50	70	30	20	
LEAD. DIS- SQL VED (UG/L AS PB)	;	:	1	<b>7</b>	æ	20	0,	71	34	<b>3</b> *	7	22	19	38	61	15	71	13	ው	82	¢	~	
TRONS OIS SOLVED (UGZL	110	140	220	, 100	180	140	92	150	100	30	0 7	120	240	90	140	140	130	160	230	200	80	270	1
COPPER, DIS- SOLVED (UG/L AS CU)		0	01	10	0	0	0	0	0	0	Э	9	0	ю	ø	ø	<b>3</b> 0	£.	~	٣	-	Vn	,
HOS- ORUS- OTAL WGAL S, PJ	1976 .030	040	100	.040	.130	.060	.070	•030	090.	•050	.080	.010	.020	.060	• 050	040	090•	090.	.050	.100	.030	.010	
	_	10. JUN	23		11 SEP	21 0CT	27.	22 DEC	:.	25. FFR	7.3. MAR	29 APR	19	11.	22 JUL	25 AUG	23 SEP	21 0CT	>02	10	22 APR	19.	

Table 11.--Water-quality analyses.--continued

Site 7, East Branch Silvermine River near Ridgefield, Conn.

DATE	TIME	FL INS TAN	REAM- OW, STAN- IEOUS FS)	SPE- CIFIC CON- DUCT- ANCE (MICRO- MHOS)	PH (UNITS)	A1	MPER- TURE ( MIR (G C)	TEMPER- ATURE: WATER (DEG C)	. 8 I	ID- TY S	YGEN. DIS- OLVED MG/L)	OXYGEN, DIS- SOLVED (PER- CENT SATUR- ATION)	COLI- FORM, TOTAL, IMMED. (COLS. PER 100 ML)
APR 13•••	1005		2.0	239	7.7		13.0	6.0	)	1	13.3	106	180
MAY 11	1010		2.3	209	7.5		21.0	13.5	j.	1	10.2	97	6500
JUN 22 23	1005 1005		1.0	238	7.4		24.5	18.0			7.9	83	2100
DATE	COLI- FORM. FECAL 0.45 UM-MF (COLS./	100 FE (CO	REP- COCCI CCAL, DLS. PER	HARD- NESS (MG/L AS CACO3)	HARD- NESS; NONCAR- BONATE (MG/L CACO3)	D) S(	LCIUM (S= DLVED (G/L G CA)	MAGNE- SIUM DIS- SOLVEL (MG/L AS MG)	SOD DI SOL (M	IUM, S- S VED	ODIUM AD- ORP- TION ATIO	BICAR- BONATE (MG/L AS HCO3)	CAR- BONATE (MG/L AS CO3)
APR 13	1	l	4	74	20		18	7.1	l	10	•5	66	0
MAY 11	12	2	95	73	13		18	6.7	·	10	•5	72	0
JUN 22 23	200		 B140	95 	16		23	9.0		11	•5	96 	0
DA	L I i	_KA- NITY NG/L NS NC03)	CARE DIOX: DIS SOLV (MG,	IDE RIC S- DIS VED SOL VL (MG	.O- RES DE, AT G DE .VED D	IDS. IDUE 180 G. C IS- LVED G/L)	SOLII DIS SOLI (TOI PEI AC-F	5- ( VED 50 NS (1 R F	IDS. DIS- DIVED TONS PER DAY)	SOLIDS. MESIDUE AT 105 DEG. C. TOTAL (MG/L)		N: MONI NO3 ORGA AL TOI /L (MO	AM- A + NIC AL
	•••	54	â	2.1	19	128		.17	.69	141		.60	.23
MAY 11 JUN	•••	59	:	3.6	17	124	,	.17	.77	122	!	•40	.38
22	•••	79 	(	6.1	19 	108	,	.15 	•29 	162		•48 	.35
DA	T(	ITRO- GEN+ DTAL 4G/L 5 N)	NITI GEI TOTI (MG,	N. PHOF AL TO /L (MC	RUS, DI FAL SO 5/L (U	PER, S- ELVED IG/L CU)	IROI DIS SOL (UG, AS (	N• NB S- ( VED S( /L (1	ANGA- LSE + DIS- DLVED JG/L S MN)	ZINC, DIS- SULVED (UG/L AS ZN)	CARB ORGA TOT (MG AS	LE ON, BL NIC ACT AL SU /L ST/	HY- NE UE IVE NCE
APR 13	•••	.83	3	•7	020	10		170	100	0	·	2.1	.00
	•••	.78	3	•5	030	0		200	120	0		3.6	.00
	•••	.83	3	•7	.050	0	;	250	140	10		8.3	.00
23							47						

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## Table 11.--Water-quality analyses.--continued Site 8, East Branch Silvermine River near North Wilton, Conn.

DATE	ΤΙ	мЕ	STRE FLO INST TANE (CF	N. An-	SPE- CIFI CON- DUCT ANCE (MICR MHOS	c - o-	PH INITS)	A T	MPER- TURE • (IR (G C)	TEMPE ATUR WATE (DEG	E, R	TUR- 810- 1TY (JTU)	OXYGEN, DIS- SOLVED (MG/L)	CEN SATU	- ED - T R-	COLI- FORM: TOTAL: IMMED: (COLS: PER 100 ML)
APR 13	11	30		6.2	1	85	7.9		12.0	8	• 0	1	12.6	. 1	05	190
MAY 11	11	50		4.9	1	65	7.9		20.0	15	•5	1	10.6		05	3400
JUN 22 23		05 15		4.2		75 	7.8		26.0	21	•5	1	8.8		99	200
DATE	COL FOR	I- M. AL. 5 MF S./	STR TOCO FEC (COL PE 100	CCI AL, S.	HARD NESS (MG/ AS CACO	r R0 N0	ARD- ESS+ NCAR- NATE MG/L ACO3)	DI SO (M	CIUM S- DLVED IG/L IG CA)	MAGNI SIUI DIS SOLVI (MG/I	M• S0 - ( ED S0 L	DDIUM+ DIS- DLVED (MG/L AS NA)	SODIUM AD- SORP- TION RATIO	BICAR BONAT (MG/ AS HCO3	E L	CAR- BONATE (MG/L AS CO3)
13		4		1		56	17		14	5	•2	8.6	•5		48	0
11 JUN		9		23		53	9		13	5	• 1	8.2	•5		54	0
22		65 		100		70 	14		18		.1	8.0	.4		68 	0
DA APR 13.	ΤE	ALKA LINII (MG/ AS CACO	r Y /L	CARH DIOXI DIS SOLV (MG/ AS CO	DE - ED L	CHLO- RIDE, DIS- SOLVED (MG/L AS CL)	DI SOL (MG	DUE	SOLII DIS SOLV (TON PEF AC-F	5- /ED : NS ?	OLIDS DIS- SOLVEI (TONS PER DAY)	RESIGNATION AT 10 DEG.	DUE G 05 NO2 C, TO AL (M	TRO- G	ITRO- EN-AI ONIA RGAN TOTAI (MG/I AS N	4-
маү 11	• • •		44		•1	13		101		.14	1.34		102	.18		33
			56 		.7	14		116	•	16	1.32		119	.37		35 
D <b>A</b> `	TE ,	NITR GEN TOTA (MG/ AS N	1• \L 'L	NITR GEN TOTA (MG/ AS NO	. P L L	PHOS- HORUS, TOTAL (MG/L AS P)	COPP DIS SOL (UG AS	- VED ∕L	IRON DIS SOLV (UG/ AS F	1. ! 5- !ED !	MANGA- NESE, DIS- SOLVEI (UG/L AS MN)	ZIN( DIS SUL) (UG)	5- ORG VED TO VL (M	BON. ANIC Tal G/L	METHI LENE BLUE ACTIV SUB- STANO	E VE - CE
	• • •	•	.49	2.	2	.020		0		90	3(	)	0	3.2	. (	00
	• • •	•	51	2.	3	.030		0	3	50	_2:(	)	10	3.9	• 1	10
	•••	•	72	3.	2	.060		0	48	20	,20		20 7.5	7.1		10

Table 11.--Water-quality analyses.--continued Site 9, Thayers Brook near Silvermine, Conn.

COLI- FORM. INTAL. IMMED. (COLS. PER	130	1400	H20	CAK- BONATE (MG/L AS CO3)	0	0	0	1 + 0		n un	20	1 W W ~	0	9	0
OXYGEN. DIS- SOLVED (PEN- CENT SATUM- ATION)	76	ž	19	BICAM- BONATE (MG/L B AS HC03) A	12	0.1	11	NITHO- GEN+DM- MONIA + OWGANIC TOTAL (MG/L AS N)	•	54.	87.	MFTHY- LENE BLOE ACTIVE SUM- STANCE (MG/L)	00.	00.	00.
O	0	•	ac.	-	٥	<b>5</b> 0	ç	NITPO- 6EN+ NO2+NU3 TOTAL (MC/L	ļ	.02	• 08	CAMBOW. OMGANIC TOTAL (MG/L AS C)	3.8	3.3	<b>9</b>
UXYGEN• DIS- SOLVED (MG/L)	11.0	9.6	γς • 3	SODIUM AD- SORP- TION HATIO	•	•	•				90		10	20	0.1
708- 810- 117 (JTU)	~	-	-	SUDIUM. UIS- SOLVEU (MG/L AS NA)	6.1	2,5	Ü.	SOLIDS. WESIDUE AT 105 UEG. C. TUTAL (MG/L)	,			ZINC. DIS- SOLVEU (UGZL AS ZN)			
		١'n			_	١.		SOLIDS. DIS- SOLVED (TONS PER DAY)	ć	1.04	.07	MANGA- NESE. DIS- SOLVEU (UG/L AS MN)	4	0 <b>6</b>	120
TEMPER- ATURE, WATER (DEG C)	10.0	16.9	23.0	MAGNE- SIUM, DIS- SOLVED (MG/L AS MG)	1.3	1.5	1.6		,		<b>a</b> o		09	0 †	0
	0.	ð. 54	5.0	ALCIUM DIS- SOLVED (MG/L AS CA)	5.0	5.	6	SOLIDS. DIS- SOLVED (TONS PEH AC-FT)	•	•	•	JHON. DIS- SOLVED (UG/L AS FE)	•	7.	49 440
TEMPER- ATUME - AIR (DEG C)	~	~	, N	O				SULIDS. WESIDUE AT 180 DEG. C DIS- SOLVED (MG/L)		y 3	19	OPPER. DIS- SSCLVED (UG/L AS CU)	c	c	၁
Hd (STINU)	4.4	6.3	6.1	HARU- NESS+ NONCAR- BONATE (MG/L CACOJ)	æ	0.1	7			~		J	0	0	2
	9	11	8 S	_	18	18	21	CHLO- RIDE. DIS- SOLVEU (MG/L AS CL)	:	. 6	12	PHOS- PHORUS. TUTAL (MG/L AS P)	070.	.040	.050
SPE- CIFIC CON- DUCT- ANCE (MICRO- MHOS)				HAMD- NESS (MG/L AS CACO3)				CARHON DIOXIDE DIS- SOLVED (MG/L AS CO2)	,		25		1.3	2.1	۲•۶
STHEAM- FLOW• INSTAN- TANEOUS		7.1	.41	STREP- TOCOCCI FECAL: (COLS: PER 100 ML)	'n	1,00	9					٩			
	0	0	0		က	96	7.2	ALKA- LINITY (MG/L AS CACO3)	,	9 60	4	NITRO- GFN. TOTAL (MG/L AS N)	. 30	74.	• 56
T I ME	1000	1010	1040	COLI- FORM. FECAL. 0.45 UM-MF (COLS./				٠,		: :	:		:	:	23
DATE	APR 14	MAY 12	23	DATE	APR 14	MAY 12	JUN 23	DATE	APR	MAY 12.	23.	DATE	APR 14	12	53

Table 11. --Water-quality analyses. -- continued

	mine, conn.
	SILVER
	near
7	ne River ne
	SILVERMI
5	orce in,
4	21.0

	COLI- FORM. TOTAL. IMMED. (COLS. PER.	4.8	2000	9150	CAR- BONATE (MG/L AS CO3)	0	0	0								
		=	66	95 t		61	21	54	NITRO- GEN•AM- MONIA • ORGAVIC TOTAL (MG/L AS v)			۰ ۵۵	METHY- LE'SE RLUE ACTIVE SIN- STANCE (MG/L)	00.	00.	• 00
	OXYGEN+ OIS- SOLVED (PER- CENT SATUR- ATION)	101	3	3	BICAR-BONATE (MG/L AS	7	N	N			.33	35.		5.6	2.4	4.3
	OXYGEN. DIS- SOLVED (MG/L)	11.4	5.5	3.	SOUTUR AD- SUMP- TION	9.	•	¢	z		o -		CARHON. UHGANIC J TOTAL (MG/L AS C)	2		
-	10x- BIO- 11 Y (JTU)	7	~	~		7.7	6.6	7.6	SOLIDS. RESIDUE AT 105 DEG. C. TOTAL		7 7	83	ZINC. DIS- SOLVEJ (UG/L AS ZN)	0.1	20	0.1
ברווודוום		0	٧,	vs	0) UI	,	2.5	10	SOL 10S+ D1S- SOL VEU (TUNS PEH UAY)		3.51	86.	MANGA- NESE. DIS- SOLVED (UG/L AS MN)	30	07	10
110	TEMPER- ATURE, WATER (DEG C)	10.0	15.5	22.5	MAGNE- SIUM. DIS- SOLVED (MG/L AS MG)	7.2	2.	8.5	•,		60.	.11	0.0	ю.	120	140
100	TEMPER- ATURE, AIR (DEG C)	21.0	17.0	25.5	CALCIUM DIS- SOLVED (MG/L AS CA)	7.0	7.1	2.5	0,			•		۰	c	۰
711	PH	7.3	7,3	1.2		1.7	=	16	SOLIDS. RESIDUE AT 180 DEG. C DIS- SOLVED (MG/L)		10	ř	COPPER. DIS- SOLVE() (UG/L AS CU)	_		
12417		20	7	•	2 11	34	87	35	CHLO- RIDE. OIS- SOLVED (MG/L AS CL)		13 4.4	1.	PHUS- PHORUS. TOTAL (MG/L AS P)	.020	•030	.040
101 00	SPE- CIFIC CON- DUCI- ANCE (MICHU- MHOS)	126	103	116	MAMD- NESS (MG/L AS CACO3)	7	~	ED.			1.S 1./	2.4		۳.	Z•8	2.3
<b>i</b> l	STREAM- FLOW+ INSTAN- TANEOUS (CFS)	2	74	4.6	STREP- TOCOCCI FECAL• (COLS• PER 100 ML)	2	480	120	CARBON DIOXIDE UIS- SULVED (MG/L AS CO2)				NITHO- GEN. TOTAL (MG/L AS NO3)	Ň		
		50	20	35		15	330	33	ALKA- LINITY (MG/L AS CACU3)		16	0≥	NITHO-GEN. TOTAL (MG/L	.51	.63	• 52
	TIME	1120	1150	1135	COL 1- FORM. FCCAL. 0.45 UM-MF (COL S./				L DATE	οχ	14 MAY 12	JUN 73	UATE	APH 14	12	33
	DATE	APR 14	MAY 12	JUN 23	DATE	APR 14	12	23	٥	APR	- ¥ -	₹~	3	A T	¥ ~ ;	2 ~

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